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A review of biocompatible polymer-functionalized two-dimensional materials: Emerging contenders for biosensors and bioelectronics applications

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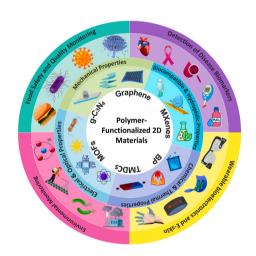
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HIGHLIGHTS

• Improved properties of polymer-2D materials for biosensing/bioelectronics.

- Discussion of diverse range of polymerfunctionalized 2D materials as biosensors.
- Insight analysis of their synthesis methods and applications.
- Critical overview of challenges and future aspects of polymer-2D materials.

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Bioelectronics, a field pivotal in monitoring and stimulating biological processes, demands innovative nanomaterials as detection platforms. Two-dimensional (2D) materials, with their thin structures and exceptional physicochemical properties, have emerged as critical substances in this research. However, these materials face challenges in biomedical applications due to issues related to their biological compatibility, adaptability, functionality, and nano-bio surface characteristics. This review examines surface modifications using covalent and

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non-covalent-based polymer-functionalization strategies to overcome these limitations by enhancing the biological compatibility, adaptability, and functionality of 2D nanomaterials. These surface modifications aim to create stable and long-lasting therapeutic effects, significantly paving the way for the practical application of polymer-functionalized 2D materials in biosensors and bioelectronics. The review paper critically summarizes the surface functionalization of 2D nanomaterials with biocompatible polymers, including $g-C_3N_4$, graphene family, MXene, BP, MOF, and TMDCs, highlighting their current state, physicochemical structures, synthesis methods, material characteristics, and applications in biosensors and bioelectronics. The paper concludes with a discussion of prospects, challenges, and numerous opportunities in the evolving field of bioelectronics.

1. Introduction

In recent years, the intersection of biotechnology, nanotechnology, and materials science has led to remarkable advancements in the development of biosensors and bioelectronics for various electronic and biomedical applications [1]. Since Luigi Galvani uncovered the concept of 'bioelectricity' in the 1780s, bioelectronics has emerged as a dynamic area of study involving the integration of electronics with biology and biotechnology [2]. Bioelectronics integrates biologically derived sensing components like DNA, cells, proteins, and antibodies with electronic components such as electrodes, field-effect transistors (FETs), optical resonators, and electrode arrays to create functional biosensors with exceptional sensitivity and specificity [3–5], as depicted in Fig. 1. These

biosensor tools are designed to identify and observe biological interactions and physiological signals. They achieve this identification through electrical or optical measurements, capturing changes in parameters such as potential, current, conductance, or resonant frequency [6,7]. With successful applications across many domains, including pharmaceutical testing, clinical diagnosis, environmental surveillance, and food quality management, these biosensors and bioelectronics devices offer viable alternatives to traditional chemical and physical sensors as well as methods [8,9].

Despite substantial success in various healthcare applications, current bioelectronic devices face limitations in spatial resolution, biocompatibility, and sensitivity due to mechanical properties and size discrepancies compared to the biological elements they aim to detect.

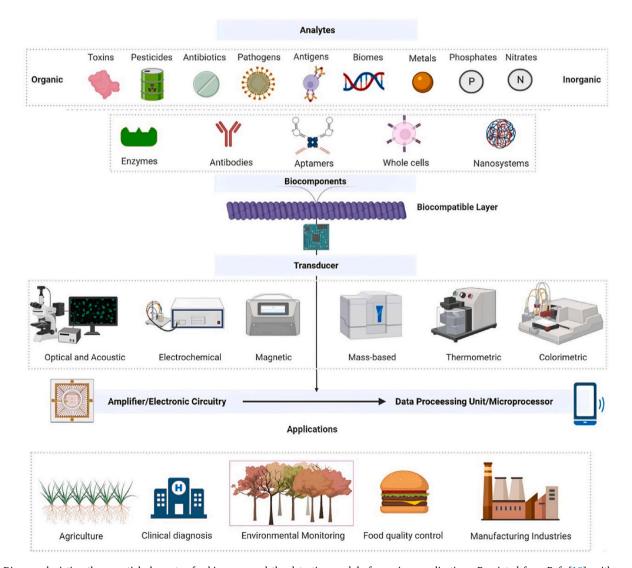


Fig. 1. Diagram depicting the essential elements of a biosensor and the detection module for various applications. Reprinted from Ref. [18]. with permission from Elsevier.

For instance, essential components involved in neuronal communication (e.g., synapses and ion channels) operate on a nanometre scale relevant to intercellular communication. Nano-electronic devices offer a means of building effective interfaces for cellular communication, serving as foundational units for seamlessly integrating electronics with biological systems [10]. However, conventional devices often differ significantly in rigidity from natural tissue, resulting in shear motion at tissue-electronic interfaces that deteriorates stimulation and recording capabilities with time. Matching electronic tools' structural, mechanical, and biochemical properties with natural tissue could enable stable integration and create new scientific and technological possibilities [11,12].

Material compatibility with biological systems is also crucial for device stability, recording reliability, and potential health risks. Investigating the impact of materials on cells and tissues before their use is essential [13]. For instance, silicon-based electronics have demonstrated utility in various biochemical and biological monitoring technologies, such as bio-photoelectric imaging tubes, biosensors, and biochips [14]. Despite their proven utility, these electronics encounter challenges like decreased carrier mobility, limited flexibility, and high short-channel properties in sub-10 nm nodes. Specifically, the quest for ultra-thin, rigid silicon-based components to enhance flexibility poses significant hurdles in integration and processing. Recently, nanomaterials and their composites have emerged, offering exceptional characteristics in combination with low modulus to create top-performing photoelectric and electronic devices capable of tolerating substantial mechanical deformation [13,15].

The primary challenge in developing these nano-biosensors lies in their capacity to reliably identify specific biomolecules or target analytes for disease diagnostics and medical interventions [16]. Consequently, when designing nano-bioelectrodes as sensing platforms, the utmost consideration revolves around achieving high selectivity and sensitivity [17]. The latest biosensors use advanced nanomaterials as sensing platforms and incorporate efficient data processing techniques for various healthcare uses. These biosensors aim to balance efficiency, stability, and multifunctionality by integrating them with state-of-the-art technologies. Overcoming their main challenges involves designing complex functional nanoplatforms and controlling their physiochemical properties [18]. Since 2004, Andre Geim and Konstantin Novoselov have successfully isolated one layer of graphene from monolithic graphite through the micromechanical cleavage method at the University of Manchester. Noteworthy attention has been directed towards nanomaterials in a two-dimensional (2D) form [19, 20]. These ultra-thin atomic layers, around 5-10 nm thick or a small number of thin layers linked together by van der Waals forces, offer broader possibilities [21]. The discovery of graphene has spurred the exploration of various 2D layered materials with remarkable mechanical, electronic, optical, and biological attributes [22-24]. Collaborative endeavours of chemists, physicists, and material scientists have explored a variety of 2D nanocrystals, encompassing transition metal dichalcogenides (TMDCs) [25], MXenes [26,27], graphitic carbon nitride (g-C₃N₄) [28,29], transition metal oxide [30], and black phosphorus (BP) [31,32] with diverse applications, including biomedicine. Beyond their 2D structure and ultra-thin nature, these emerging materials showcase distinct electronic, optical, and physicochemical characteristics like efficient carrier transfer, extensive surface areas, transparency, biocompatibility, flexible layered structures, robust interfacial attractions with cells, DNA, biomolecules, proteins, and other biological entities [33,34]. Consequently, these distinctive attributes make them suitable not only for bioelectronic and biosensor devices but also for overcoming the limitations of silicon-based electronics [35].

Despite the impressive strides made in developing bioelectronic devices combined with nanotechnology, they still need to match the effectiveness of biological systems in performing tasks like molecular recognition, catalyzing chemical reactions, and reacting and adapting to changes in their surroundings. As a result, the collaboration between 2D materials and polymers has led to significant progress in bio-interfaces,

offering exciting prospects for regulating biomolecular interactions within bioelectrodes and mimicking natural environments to enhance electronic capabilities [36]. Polymers, known for their adaptability, biocompatibility, and ease of synthesis, are frequently employed in surface functionalization. The ultrathin structures and extensive surface areas of 2D materials make incorporating various elements, like polymers. This synergy between polymers and 2D materials has driven the development of innovative nanodevices that can be processed with solvents, stable monitoring platforms, and adhesive nanointerfaces [15]. Challenges like agglomeration in 2D materials can also be addressed by blending them with polymers in a thickened solution, where polymers reduce water molecule flow abilities and slow diffusion of 2D materials, contributing to stability and optimal performance [37].

To safely use 2D nanomaterials in medical applications, the primary concern is to ensure their safety in biosystems, including in vitro and in vivo diagnostic scenarios. It has been observed that the chemical composition influences the biocompatibility of 2D materials, and surface modification with polymeric substances like bovine serum albumin (BSA), poly(vinyl pyrrolidone) (PVP), polyethylene glycol (PEG), etc., can enhance their stability, biocompatibility, and physiological dispersibility [38]. For instance, MoS₂ nanosheets (NSs) exfoliated with BSA demonstrated better compatibility with fibroblast cells compared to those without BSA [39]. Such surface modifications are commonly employed to stabilize 2D materials under physiological conditions, improving biological compatibility despite potential toxicity concerns.

There has been a significant surge in interest within the biosensing and bioelectronics fields in recent years, with numerous published studies. However, previous research has predominantly concentrated on specific subsets of this broad field, often examining individual classes of 2D materials for singular applications [40,41] or a single composite of polymer-functionalized 2D materials within a particular domain of biosensing [42]. For instance, Jiang and his team focused on the development, mechanisms, challenges, and future perspectives of graphene-based biosensors for detecting bacterial and viral pathogens [43]. In another review, Alwarappan and colleagues comprehensively discussed the synthesis routes, surface functionalization, and application of MXenes in biosensors [16]. Other researchers have broadly covered recent trends in the design and fabrication of graphene/polymer nanocomposites (GPNs) for sensing applications in environmental and human health monitoring [44]. However, our study broadens the scope of research by exploring the functionalization of various 2D materials with different polymers and covering a wide range of real-life applications of these materials in biosensors. This broad scope is not comprehensively covered by other relative studies with the latest research from 2014 to 2024. The review begins with an overview of the synthesis methods for 2D materials, emphasizing those functionalized with polymers. We then elaborate on the distinctive characteristics that make these materials highly compatible with bioelectronics and biosensors. The discussion will comprehensively address how specific polymer properties enhance the biosensing capabilities of various emerging 2D materials. Furthermore, we outline these composite materials' sensitivity and detection limits in particular applications, illustrating their practical implementations. The applications covered include food safety, environmental monitoring, detection of disease biomarkers, and the development of wearable bioelectronics and electronic skins (E-skins). Finally, the study explores the challenges and prospects of applying functionalized 2D materials in bioelectronics and biosensors, providing insights into future research and development directions in this rapidly evolving field.

2. Synthesis of polymer-functionalized 2D materials

The synthesis of 2D nanomaterials and their derivatives with polymers can be classified into two main categories: top-down and bottom-up approaches. The bottom-up approach can be further subdivided into various methods, including ligand-mediated growth, small molecules

assisted bottom-up approach, polyhydric alcohol method, and hydrothermal method, among others. On the other hand, the top-down method encompasses techniques such as exfoliation methods (such as chemical, micromechanical, ultrasonic, tiny atom assisted, and liquid phase exfoliation), nanolithography, laser ablation, and more [45]. Some commonly used synthesis methods are outlined in this review paper section.

2.1. Chemical vapor deposition

Chemical vapor deposition (CVD) is a high-temperature technique to vaporize and deposit metal on a suitable substrate. This method allows for synthesizing 2D nanomaterials based on transition metals, which provide superior electrical conductivity and more control over the final nanosheet or nanoplate thickness. CVD can create nanosheets in metal, semiconductors, and insulators. High temperatures evaporate the material in an adequate substrate and an inert environment; a tube furnace usually aids this process. The outer tube introduces the precursor and the inert atmosphere. The ability to carefully regulate the qualities incorporated into the nanomaterials, such as shape and crystallinity, is a crucial benefit of the CVD process. This approach allows the synthesis of high-purity 2D materials. CVD is one of the most popular techniques for creating 2D nanomaterials, including graphene, TMDs, and other layered materials. The basic idea is that a precursor gas breaks down on a substrate to form a thin film or 2D substance [45].

Utilizing the expansive surface area and superior quality of 2D materials, particularly those synthesized via CVD, offers a straightforward approach to precisely tailoring the surface or interface within layers of polymer structures. Typically, polymers like thermoplastic poly (methyl methacrylate) (PMMA), polystyrene (PS), polycarbonate (PC), polyolefins, and other compatible materials are employed as support mediums. The 2D sheets can be easily transferred from their growth substrates, such as copper foil or silica, onto the designated polymer support. This process enables the creation of composite films with a 2D sheet serving as the surface layer, which can then be utilized as precursors or foundational elements for subsequent processes, such as the fabrication of nanocomposites or devices [46].

2.2. Liquid phase exfoliation

Among the array of available exfoliation methods, liquid-phase exfoliation (LPE) is the most widely acclaimed technique for producing diverse types of 2D materials, offering precise control over their dimensions by adjusting process parameters. LPE typically involves either sonication-assisted or high-shear exfoliation. Initially, the bulk materials are dispersed in a suitable solvent with or without the aid of a surfactant and then subjected to either sonication or high shear mixing. Throughout the LPE process, the growth and subsequent collapse of micrometer sized bubbles or voids exert forces on the bulk material, effectively separating it into individual sheets [47]. Additionally, surfactants can improve the stability of the exfoliated sheets by intercalating between the layers, preventing their tendency to re-aggregate. LPE is typically conducted under mild conditions without high temperature or pressure. Offering high yield and adjustable parameters, 2D materials acquired via LPE find versatile applications, particularly in the biomedical realm [48,49].

Exfoliating agents that effectively produce water-soluble and functional nanosheets have been found within polymers known for their favourable biocompatibility. PVP, distinguished by its non-ionic and non-toxic properties, has emerged as a widely utilized exfoliating agent. The hydrophilic properties of PVP and substituents present on the nitrogen atom of the pyrrolidone ring endow its structure with similarities to N-methyl pyrrolidone (NMP), thereby contributing to the improved colloidal stability of aqueous nanosheet suspensions. Additionally, nanosheets exfoliated with PVP exhibit commendable biocompatibility, rendering them suitable for diverse bio-applications [50]. Nevertheless,

various LPE methodologies and chemical functionalization techniques may influence the biocompatibility of the resultant materials. Hence, evaluating potential adverse effects, such as toxicity, inflammation, or immune responses, is crucial when utilizing 2D materials to ensure their suitability for clinical applications [48,49].

2.3. Micromechanical exfoliation

Atomically thin sheets of diverse 2D inorganic materials can be acquired through micromechanical exfoliation, a method like the one Grim and his colleagues employed for isolating single-layer graphene from highly oriented pyrolytic graphite (HOPG). This technique involves delicately rubbing or peeling commercially available inorganic layers of material against scotch tape on a photoresist surface. As a result, various flakes, including monolayers, adhere to the photoresist surface due to weakening van der Waals forces between adjacent layers. Subsequently, these monolayers are rinsed with acetone and transferred onto an oxidized silicon wafer. Although this method facilitates the extraction of monolayers from bulk materials like WS2, MoS2, h-BH, and NbSe₂ in a relatively straightforward, rapid, and cost-effective manner, it does come with certain drawbacks. Since monolayers serve as the foundational component of layered materials, separating them from thicker flakes is crucial for understanding the intrinsic properties of these materials. However, the yield of monolayers from this process is notably low, limiting its applicability primarily to laboratory-scale investigations rather than large-scale production for advanced technological applications. Nevertheless, the absence of chemical involvement in this process ensures structural integrity and high crystallinity maintenance, as it relies solely on applying shear force during the peeling procedure [51]. This technique has the potential to synthesize various 2D materials, as demonstrated by Gomez and his team, who utilized a polydimethylsiloxane (PDMS)-based intermediate viscoelastic substrate to achieve high-yield production of atomically thin BP flakes [52].

2.4. Hydrothermal method

Also known as the solvothermal method, this method operates under high temperature and pressure conditions, using water as an aqueous solvent or an organic solvent alongside NaOH as a reducing agent. The reaction mixture is enclosed within a sealed vessel made of Teflon encased in a thick stainless-steel chamber, functioning as an autoclave. Organic solvents are utilized in the solvothermal approach. In the midnineteenth century, geologists sought to replicate hydrothermal conditions to mimic the formation of specific rocks and minerals, hence coining the term "hydrothermal." Granular production has existed for two centuries, based on hydrothermal techniques for single-crystal formation. This method employs a hydrophilic solution within a distinct sealed reaction vessel to create an environment of elevated temperature and pressure by heating and pressurizing the reaction system. This enables compounds that are typically poorly soluble or insoluble under standard conditions to dissolve and recrystallize at high temperatures and pressures [45].

Hydrothermal (or solvothermal) synthesis has garnered considerable attention in 2D material synthesis. Like liquid-exfoliated 2D materials, the hydrothermal method facilitates the synthesis of -O or -OH groupfunctionalized 2D materials using water as a solvent during the synthesis process. Additionally, incorporating groups other than OH is feasible with carefully selected additives for the reaction. For instance, the synthesis of amino-functionalized graphene and polymerfunctionalized MoS $_2$ using additives such as PEG and PVP has been recognized. Hydrothermal synthesis is particularly adept at producing small particles rather than large and thin sheets [53].

3. Material properties required for biosensors and bioelectronics

2D materials offer distinctive properties that make them compatible with bioelectronics and biosensors, yet they face challenges like poor adaptability, cytotoxicity, low charge carrier mobility, environmental degradation, poor light absorption, and durability. Surface functionalization has emerged as a well-explored and practical strategy to address these challenges. This method allows for manipulating 2D nanomaterial properties while preserving their original lattice structure. Consequently, this approach is ideal for creating 2D material or polymers with unique traits like improved optical absorption, environmental stability, low cytotoxicity, robust mechanical strength, and the capacity to detect chemicals and biomolecules. This section highlights the advantageous properties of these materials in biomedical and flexible/stretchable bioelectronics, highlighting the latent potential of wearable electronics as combined biosensing systems that conform to the skin.

3.1. Mechanical properties

Good mechanical properties like flexibility and durability enable conformal contact with curvilinear, dynamic human skin and make acute health measurements possible. They also play a pivotal role in the

lifespan of skin bioelectronics for long-term, continuous health monitoring. For materials used in these applications, finding the right balance in mechanical properties is essential yet challenging [54]. They need to be strong enough to withstand significant deformations while remaining flexible to match the softness of body tissues. The ability to stretch should be over 10 % on flat skins and 60 % on full-range body motions, depending on the device placement and target applications. In past years, the progressive development of 2D materials has emerged as excellent materials for bioelectronic applications because of their robust in-plane covalent or ionic bonds, strength, and atomic-thin structure, offering flexibility and stretchability [38]. However, to further improve the long-term stability, flexibility, and stretchability to ensure the viability of these materials in practical applications, scientists are actively functionalizing 2D materials with various polymer matrices to create 2D material/polymer hybrids with improved mechanical properties [55].

For instance, a $GOx/gold/MoS_2/gold$ nanofilm over a polymer electrode was fabricated to enhance glucose detection by improving electrochemical signals and flexibility [56]. Utilizing a flexible polymer electrode as a substrate, the biosensor demonstrated superior flexibility to conventional Au-coated silicon sensors, as measured by a micro-fatigue tester (Fig. 2A and B). The Au-coated silicon electrode showed a flexure extension value of just 0.0938 mm under a force of

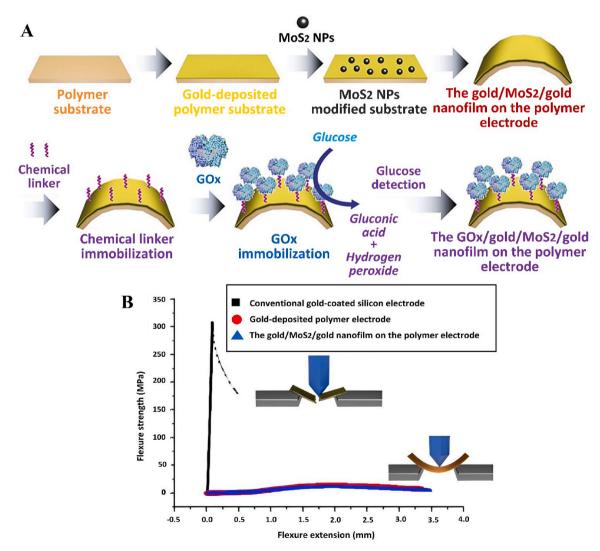


Fig. 2. A) Schematic representation of the construction process of a flexible electrochemical glucose biosensor featuring a GOx/gold/MoS₂/gold nanofilm on a polymer electrode. **B)** Demonstrating the flexibility of nanofilm on polymer electrodes through a flexibility test. Reproduced from Ref. [56]. with permission from *Elsevier*. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

307.4 MPa. In contrast, the gold/MoS₂/gold nanofilm over an electrode of polymer and Au-deposited polymer showed values of 3.48 mm and 3.36 mm, respectively, under approximately 12 MPa force, highlighting their flexibility. The electrochemical analysis demonstrated an increased signal with the nanofilm compared to conventional electrodes, indicating enhanced sensitivity. Aerometric measurements revealed the biosensor's ability to detect glucose, with a limit of detection (LOD) of 10 nM in a linear range of 500-10 nM and excellent selectivity even in human serum samples. In conclusion, the flexible glucose biosensor developed in this study showed promising potential for wearable biosensing systems due to its enhanced sensitivity, selectivity, and flexibility. Further optimization may be required for practical applications, particularly ensuring uniform layer depth. Wang and coworkers fabricated PP-rGO20 nanocomposites onto microelectrode sites, eliminating the need for post-reduction for neutral interface use [57]. TEM images revealed the polymerization of PEDOT: PSS on the rGO sheet surface. This approach unveiled the internal structure of PP-rGO20 and demonstrated the attachment of PSS to rGO sheets through π - π attractions between PSS and the sp² electron-stacking of rGO. The composite enhanced microelectrodes' stability and electrochemical properties and demonstrated robust mechanical stability under cyclic voltammetry (CV) scanning and ultrasonication tests. After 1000 cycles of scanning (n = 3), a 29.8 \pm 5.1 % decrease in charge storage capacity (CSC) and an 11.6 ± 3.0 % increase in impedance at 1 kHz confirmed the superior mechanical stability of the PP-rGO20-modified electrode during stimulation and recording, even during in-vivo acute implantation. Yet, the study lacks long-term biocompatibility assessment and stability under extended electrical stimulation, highlighting the need for comprehensive evaluation and optimization for practical neural interface applications.

3.2. Electrical and optical properties

Optical and electrical properties depending on band gaps and electronic structures of 2D materials are crucial for biosensing applications [18]. The crystal structure of 2D materials affects photon and electron movement, with atomic thickness offering optical transparency but reducing charge mobility due to electron scattering. For instance, graphene's 0 eV band gap limits its use in optoelectronics for biosystems. Yet, its atomic thickness and high carrier mobility offer high optical transparency and fast response in optoelectronics. Many 2D materials have band gaps and electrical mobility dependent on the thickness of the layers, like MoS₂ transitions from an indirect to a direct band gap with varying numbers of layers. Additionally, BP has a thickness-dependent band gap (0.2-2.1 eV) and high carrier mobility at 10 nm layer thickness. Due to their relative band gaps, these 2D materials, with their more significant absorption coefficients at distinct wavelengths than graphene, are more suitable for optoelectronic devices [38]. However, the unique optical features of graphene-based nanomaterials, including tuneable and broadband absorption, make them valuable for optical sensors in various applications [58]. The bandgap and electrical properties of graphene and other 2D materials can be improved through different means, like hybridization with other nanomaterials, doping, or surface treatments. It is suggested that the optoelectronic properties of 2D nanomaterials can be significantly enhanced through surface functionalization with suitable polymers, opening new possibilities for applications in biosensors and bioelectronics.

Deviations in optical properties could be utilized for biomarker detection, as demonstrated by Zhou and his team, who developed a fibre optic biosensor built on 2D BP for sensing the cancer biomarker called human neuron-specific enolase (NSE) [59]. The nanosheets of BP were bio-functionalized with poly-L-lysine (PLL), serving as a crucial cross-linker to enhance the bio-nano-photonic interface with a significantly improved light-matter interface. Immobilizing the anti-NSE antibody through PLL allowed for the real-time and susceptible detection of NSE, as the linking of NSE by antibody altered the local refractive

index. Due to its remarkable performance, they believe this BP-based biosensor opens new possibilities for bio-nano-photonic applications in food safety, healthcare, biomedical, and environmental monitoring. Future work should evaluate the biosensor's performance with clinical samples to validate its practical utility in early cancer diagnosis and disease detection. Photoelectrochemistry (PEC) involves the conversion of light energy into electricity by semiconductors (photoactive material), generating electron-hole pairs that initiate redox reactions at interfaces upon illumination [60]. A PEC sensor for glutathione detection using a nanocomposite of MoS2 and polyaniline (PANI) was devised as the optoelectronic material [61]. The PANI/nanoMoS₂/GE nanocomposite exhibited a maximum PEC response of 712 nA, surpassing the performance of single material-modified gold electrodes with PANI or MoS₂. This improvement was attributed to the effective prevention of electron-hole pair recombination in PANI/nanoMoS2, leading to an improved photocurrent response. The optimization of various parameters, such as ultrasonic time (4 h), ratio of PANI to nanoMoS₂ (5:2), and pH (6.0), suggested that the biosensor's performance was sensitive to experimental conditions. Achieving optimal results may require extensive experimentation and fine-tuning, which could increase the time and resources necessary for sensor development.

The enhanced electrical conductivity promotes efficient electron transmission between the solution and the modified electrode. A straightforward co-deposition technique anchors a PEDOT/g-C₃N₄ hybrid film on GCE. The g-C₃N₄ incorporation into PEDOT addressed the low conductivity of g-C₃N₄ and significantly enhanced electron-transfer capabilities [62]. The findings suggested that the modified electrode with the composite film displayed superior electrocatalytic activity for sensing ascorbic acid (AA) and acetaminophen (AP), offering a low LOD and broad linear range. Similarly, real sample analysis using Vitamin C and Paracetamol tablets demonstrated the biosensor's practical feasibility, with satisfactory recovery rates (94.9-104.56 %) indicating reliable detection capability. Yet, addressing further studies focusing on long-term stability, comprehensive selectivity testing, and validation using a diverse range of real samples would strengthen the reliability and applicability of the PEDOT/g-C₃N₄/GCE sensor. In another study, electrochemically deposited PEDOT was utilized as the conductive framework to create a composite with g-C₃N₄, addressing the limited conductivity issue of g-C₃N₄ [63]. The strong interactions and the cooperative influence between E-PEDOT and g-C₃N₄ also greatly enhanced the electron transmission properties and improved the electrochemical catalytic activity of the nanocomposite. Compared to prisg-C3N4, and E-PEDOT-modified electrodes, g-C₃N₄-E-PEDOT-modified GCE exhibited superior electrocatalytic activity during AP oxidation. This study achieved a lower LOD for detecting AP of 34.28 nM compared to the above research with an LOD for AP of $0.49 \mu M$. Overall, while both studies showcased innovative approaches and successful applications of composite materials in electrochemical sensing, they differed in the specific composite materials used, target analytes detected, and optimization strategies employed.

3.3. Chemical and thermal properties

The stability of 2D nanomaterials in biosensors and bioelectronic applications is determined by their chemical and thermal properties. Various factors, such as bonding, structure, doping, functional groups, defects, and environmental conditions, can influence the chemical and thermal stability of 2D nanomaterials. In particular, the in-plane thermal conductivity and cross-plane heat resistance of 2D nanomaterials are noteworthy considerations due to their ionic or covalent in-plane bonds and weak van der Waals interactions between layers [38]. In sensing applications, sensors must exhibit stability in both thermal and chemical aspects because the sensing material experiences various heat and chemical treatments throughout the machine processes involved in sensor manufacturing. Even continuous health monitoring relies heavily on the enduring stability of skin electronics. Hence, considerable efforts

have been made to improve the chemical and thermal stability of 2D materials because materials like BP and MXenes are easily susceptible to degradation in ambient conditions, primarily triggered by oxygen and water exposure. For example, the stability of BP NSs is notably impacted by light, which can trigger the formation of various reactive oxygen species (ROS) in the presence of oxygen and water [55]. On the other hand, under oxidative conditions such as elevated temperatures, ${\rm Ti}_3{\rm C}_2{\rm T}_{\rm X}$ (a type of MXene) transforms ${\rm Ti}_2{\rm C}_2{\rm T}_2{\rm C}_2{\rm T}_2{\rm C}_2{\rm T}_2{\rm C}_2{\rm T}_2{\rm C}_2{\rm T}_2{\rm C}_2{\rm C}_2{\rm$

For instance, a flexible biosensor utilizing hybrid sensitive films of PANI and Ti₃C₂T_x was crafted to monitor NH₃ volatilization in agriculture [64]. The core-shell structure of Ti₃C₂T_x/PANI hybrids ensured the stability of 88 % for 35 days at ambient conditions, with outstanding humidity tolerance of up to 60 % relative humidity (RH) within 10-40 °C making them well-suited for practical applications. The superior performance of the hybrid sensor compared to pure PANI was attributed to factors like increased surface area, enhanced protonation degree of PANI, and the formation of a Ti₃C₂T_x/PANI Schottky junction. While the sensor exhibited excellent sensitivity to NH3 even in high-humidity environments, it showed increased sensitivity to water vapor, especially at high RH levels (>90 %). This high sensitivity to humidity can potentially interfere with accurate NH3 detection, particularly in environments with fluctuating humidity levels. The RGO-phenol formaldehyde composites are formed by in-situ intercalate polymerization via an interactive oxidation-reduction reaction. It involved the reduction of GO by phenol to RGO and the simultaneous oxidation of phenol to benzoquinone [65]. Here, the phenol noncovalently adsorbed on RGO was a reductant that guided the PF formation on the RGO surface. The interaction between GO and phenol homologs via covalent bonding and π - π stacking ensured strong interfacial interactions and good dispersion, thus resulting in enhanced thermal conductivity and thermal resistance. It was noted that adding RGO to polymer composites enhanced thermal conductivity and stability, particularly with 0.5 wt% RGO loading. However, higher concentrations led to decreased conductivity due to RGO agglomeration. Moreover, other thermal analyses confirmed enhanced thermal resistance and higher decomposition temperatures in RGO-PF composites, indicating their potential for high-performance thermal management applications. Yet, the study primarily focused on the structural and functional properties of RGO and its composites, with limited exploration of their practical applications or scalability.

Here, a novel cholesterol biosensor using a composite material, CSPPy-g-C₃N₄H⁺, as the matrix was introduced [66]. The composite material, formed by exfoliating g-C₃N₄H⁺ into thin nanosheets, provided an effective and large surface area. Consequently, the fabricated biosensor exhibited good long-standing storage stability over 45 days despite various potential interfering elements, retaining 94 % of its initial response. It also exhibited repeatability, with an average RSD of ≤3.46 % in ten repeated measurements, confirming excellent performance. The study does not provide a comparative analysis with existing cholesterol biosensors. A comparative evaluation could highlight the advantages and limitations of the proposed biosensor relative to existing technologies, providing valuable context for its performance. A high surface passivation and effective exfoliation of BP nanosheets using a diluted polymer ionic liquid solution were achieved for flexible photoelectrodes. The stability of the resultant PIL-modified BP was significantly enhanced in both aqueous and solid-state environments compared to bare BP [67]. Even after 120 h of exposure to ambient conditions, the modified BP's photocurrent density remained constant at approximately 50 nA cm⁻². This impressive stability of the flexible

photodetector was credited to the exceptional photostability and ambient of the BP/PIL materials. When this material was exposed to ambient conditions for 100 days, its crystalline nature remained unchanged, showcasing the excellent environmental stability conferred by PIL modification.

3.4. Hydrophilicity and biocompatibility

Skin bioelectronics and biosensor devices are known for their direct interaction with human skin, necessitating careful consideration of biocompatibility, especially for prolonged usage. Extensive work has focused on creating biocompatible active materials and encapsulating layers. This effort has resulted in various biocompatible skin bioelectronics, from sensors and energy devices to integrated circuits [9]. The toxicity of 2D materials stems from variations in synthesis and post-processing steps, altering material composition and degradation characteristics, which can significantly impact the nano-bio interface. Factors like exfoliation and lateral dimensions also influence their toxicity. Notably, for certain materials like TMDs, particularly MoS2 and BP, higher exfoliation states correlate with increased toxicity due to active edge sites and expanded surface area. Thinner and smaller 2D nanomaterials exhibit excellent toxicity responses due to their enhanced effective surface area. Additionally, material characteristics such as crystal structure, surface functionality, chemical changes, and structural form play roles in toxicity. However, the lack of comprehensive material characterization and toxicological studies hinders a precise assessment of the hazard potential of 2D nanomaterials. To address this concern, biocompatible polymers have emerged as a popular approach to enhance the biocompatibility and stability of inherently toxic 2D materials such as MoS2 and graphene [68]. Hydrophilicity, linked to biocompatibility, tends to be higher with better biocompatibility. For example, the hydrophobic nature of graphene surfaces can lead to strong interactions with biomolecules, potentially causing pore clogging. While surface modifications like oxidation and coating are possible, they may impact membrane integrity or alter pore size, affecting biosensors' signal-to-noise ratio (S/N) [69]. Hence, researchers are exploring 2D materials with more hydrophilic surfaces for nano-biosensor development.

The biocompatibility and hydrophilicity of graphene (GR) are enhanced by covalently attaching hydrophilic chitosan (CS) biopolymer while maintaining its electronic performance [70]. CS showcases remarkable traits such as superior film-forming capacity, biocompatibility, strong adhesion, non-toxicity, and potential for chemical alterations due to its abundant amino and hydroxyl groups. Leveraging these distinctive characteristics, it is logical to anticipate enhancements in the hydrophilicity and biocompatibility of GR through CS-functionalization via covalent grafting methods. It was observed that the successful grafting of CS onto GR without structural damage allowed for dense Pd NP decoration onto CS-GR sheets through in situ reduction. The resulting electrode modified by Pd NPs/CS-GR exhibited outstanding electrocatalytic activity for glucose oxidase (GOD) and H₂O₂ oxidation, holding promise for various oxidase-based bioelectronics and biosensors. However, the study primarily focused on biosensor performance under controlled laboratory conditions. Further validation in real-world scenarios, including complex sample matrices and interference studies, would be essential to assess its practical utility and reliability in clinical or industrial settings. An eco-friendly method to create PSGO-PEDOT nanosheets has been developed for their applications in bioelectronics and biomaterials [71]. PSGO-PEDOT nanosheets were made using a functional GO template for this purpose. GO underwent two sequential functionalization steps. First, sulfonate groups were added to the GO surfaces to produce sulfonated GO (SGO), providing doping sites for PEDOT. In the next step, catechol groups were introduced onto SGO, resulting in PSGO through PDA functionalization. This process enhanced interactions between PEDOT and PSGO. Furthermore, PDA also reduced PSGO, thereby increasing its conductivity. Consequently, PEDOT

assembled on the PSGO template to form conductive sandwich-like nanosheets with abundant hydrophilic and reductive catechol groups, leading to good water dispersibility and redox activity. These nanosheets then created physical crosslinks dispersed uniformly within a chemically crosslinked polyacrylamide (PAM) hydrogel network. This combined physical and chemical crosslinking imparted toughness and stretchability to the hydrogel, thanks to the synergistic effects. This hydrogel also displayed notable biocompatibility in vitro and in vivo for 14 days, presenting great promise for highly sensitive biosensor applications (Fig. 3A–D). Since the study demonstrated promising results in vitro and in vivo, further validation in diverse physiological conditions and long-term biocompatibility studies are essential to assess its suitability for clinical applications.

In another study, a covalent biofunctionalization strategy using PDA as a nano-bio interface was introduced to enhance the stability and performance of MoS₂ nanosheets for intracellular biosensing [72]. Monolayer MoS_2 nanosheets were prepared ultrasonication-assisted lithium intercalation approach, followed by PDA coating to form MoS₂@PDA nanosheets (MP Thiol-terminated polyethylene glycol (PEG-SH) was then conjugated to MP NSs to form MoS₂@PDA-PEG (MPP) NSs. Subsequently, fluorescein-labeled substrate peptide (P1) and cell-penetrating peptide (P2) were covalently conjugated to MPP NSs, forming MoS2@PDA--PEG-Peptide (MPPP) biosensors via Michael addition reactions. The MPPP biosensor demonstrated efficient caspase-3 detection, with a linear response of 2-360 ng/mL and a LOD of 0.33 ng/mL. The cytotoxicity of the biosensor was assessed on human cervix carcinoma (HeLa) cells using a lactate dehydrogenase (LDH) assay after 24 h of exposure. Results showed cell viability exceeding 95 % across 4-128 µg/mL concentrations, confirming good biocompatibility. Future studies may focus on in vivo applications and further optimizations for enhanced sensitivity and multiplexed detection capabilities.

4. Functionalized pioneering 2D materials in biosensors and bioelectronics

After the remarkable success of graphene, various newly arisen 2D materials, including $g\text{-}G_3N_4$, TMDCs, MOFs, BP, and MXene NSs, have captured considerable attention in materials science and bioelectronics. Due to their exceptional physicochemical properties, these materials, including the graphene family, can be effectively functionalized with polymers to achieve impressive qualities such as outstanding flexibility, adjustable electrical conductivity, diverse functionalities, stability, and high biocompatibility. Table 1 shows a detailed overview of each class, with a comprehensive discussion in the following sections.

4.1. 2D graphitic carbon nitrides

A non-metallic semiconductor, g-C₃N₄ in sheet form, shares similarities with graphite [73–75]. It has garnered significant attention due to its easy synthesis from C- and N-rich precursors like melamine in aqueous media [76,77] (Fig. 4A). Compared to their bulk counterparts, 2D g-C₃N₄ nanostructures offer enhanced attributes, including improved biocompatibility, adjustable bandgap, enhanced electrical conductivity, greater surface area, reduced electron-hole pair recombination rate, and superior optical absorption with high quantum yield [76,78]. These semiconductor properties, such as electrochemiluminescence (ECL), photo-electrochemistry, and photoluminescence (PL), along with their electron-donating capability, make g-C₃N₄ a promising nanoplatform for diverse applications, including hydrogen production, biosensing,

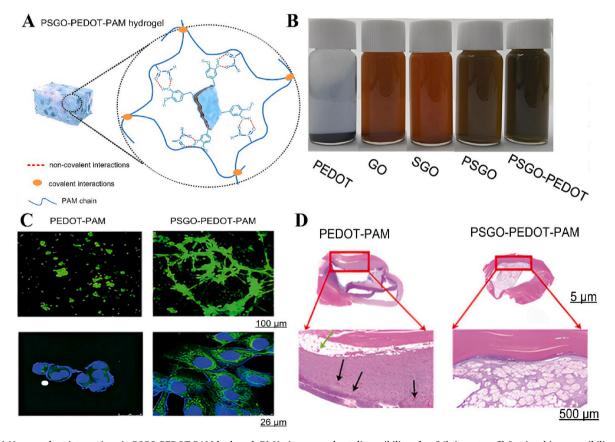


Fig. 3. A) Non-covalent interactions in PSGO-PEDOT-PAM hydrogel. B) Various nanosheet dispersibility after 24h in water. C) In vitro biocompatibility of PEDOTPAM and PSGO-PEDOT-PAM hydrogels, with cell morphologies (up) and focal adhesion staining (down). D) Histologic images of hydrogel implanted in rabbit muscle after 14 days, indicating inflammation (black arrow) and edema (green arrow). Reproduced from Ref. [71]. with permission from Wiley. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

 Table 1

 Polymer-functionalized 2D materials in biosensor design: Overview of physiochemical interaction between polymers and 2D materials, analytical performance, advantages, and challenges.

Biosensor designs	Interactions	Analytes	Linear range	LOD	Advantages	Challenges	Reference
2D Graphitic carbon r DexP- g-C ₃ N ₄	nitride π–π stacking	Con A	500–100 ng/ mL	17 pg/Ml	Excellent ECL behavior, large surface area, biocompatibility, specificity, long-term storage stability (up to 15	Experimental conditions optimization, low sensitivity	[93]
C-g-C ₃ N ₄ -PEI	Covalent bonding	OPs	0.001–5000 nM	0.3 pM	days) Simple and sensitive analysis	Low luminous efficiency and stability, AChE	[94]
g-C ₃ N ₄ /chitosan	Electrostatic attraction	Hg ⁺²	$\begin{array}{c} 1.00-80.0,\\ 0.1-5~\mu mol\\ L^{-1} \end{array}$	$\begin{array}{c} 0.01 \\ \mu mol \ L^{-1} \end{array}$	Improved mechanical, electrochemical activity, electron transfer, selective, and sensitivity	immobilization Limited linear range	[230]
2D Graphene family			-				
hemin/RGO-CMF	-	H_2O_2	0.06–540.6 μΜ	~16 nM	Cost-effective, eco-friendly, high electro-reduction ability, superior selectivity, sensitivity (0.51 μ A μ A $^{-1}$ cm $^{-2}$) at S/N = 3, rapid response (4 s)	Hemin stabilization, experimental conditions optimization	[231]
Graphene-PANI	-	HPV type 16 DNA	10–200 nM	2.3 nM	Cost-effective, high selectivity, reproducible inkjet printing, scalable mass production, enhanced	Excess Agglomeration of G- PANI layers, complicated and time-consuming covalent probe immobilization	[232]
ChOx/PEI-rGO	Electrostatic attractions	Cholesterol	$\begin{array}{l} 0.10\times 10^{-6} - \\ 9.331\times 10^{-3} \\ mol \ L^{-1} \end{array}$	$\begin{array}{c} 0.021 \\ \mu mol \; L^{-1} \end{array}$	electroactive surface area Excellent dispersion of PEI-r GO in water, biocompatible, increased enzyme immobilization, direct electron transfer, anti-interference ability, stability (99 % over 50 cyclic scans)	Optimization of fabrication conditions, electrode stability	[233]
Graphene-Nafion	Physical	HIV-1	$1 \times 10^{-13} - 1$	2.3 ×	Simple method, label-free, high	Limited LOD detection	[114]
DANI /graphene	adsorption Electrostatic	HIV-1	$ imes 10^{-10} \ ext{M} \\ 5.0 imes 10^{-16} \ ext{-}$	10^{-14} M $1.0 \times$	sensitivity, specific surface area High sensitivity, electrical	DNA immobilization,	[111]
PANI/graphene	attraction	HIV-I	1.0×10^{-10}	1.0 × 10 ⁻¹⁶ M	conductivity, selectivity, and stability	specificity, sensitivity optimization	[111]
rGO/PEI	Electrostatic attraction	E. coli	1.0×10^{1} – 1.0 × 10^{4} cfu mL ⁻¹	~10 cfu mL ⁻¹	Sensitive and selective detection in aqueous and serum samples	Adaptability, specificity optimization, matrix variability in real samples	[118]
2D MXenes nanostruc	ture					0 . 0 . 1 . 1	
TiO₂(G) NW@EPNS/Cyt C	_	NO ₂	0.5–9000 mM	0.225 mM	Enhanced photocatalytic oxidation, electron transfer, sensitivity (201.79 $\mu A \mu M^{-1} \text{ cm}^{-2}$), selectivity, reproducibility, biocompatibility,	Cyt C stability in TiO ₂ (G) NW@EPNS matrix, optimization of detection conditions, EPNS stability	[117]
PPy@Ti ₃ C ₂ T _x / PMo ₁₂	-	OPN biomarker	500–10 ng/mL	$\begin{array}{c} 0.98 \text{ fg} \\ \text{mL}^{-1} \end{array}$	and reusability Strong binding ability towards RNA aptamer, high sensitivity, selectivity, stability, and reusability	Hindered electron transfer, complex fabrication procedure	[136]
GOx/Au/Ti ₃ C ₂ / Nafion/GCE	-	Glucose	0.1 mM-18 mM	5.9 μΜ	Improved electron transfer kinetics, selectivity, sensitivity (4.2 μ AmM -1 cm -2) at S/N = 3, and storage stability	Uniform and stable film coating on GCE	[133]
CIPs/AuNPs/ Ti ₃ C ₂ T _x /GCE	-	Yeast cells	$\begin{array}{l} 1\times10^21\times\\ 10^9 \text{ cells per}\\ \text{mL} \end{array}$	20 cells per mL	Enhanced electron transfer rate, surface area, sensitivity, selectivity, reproducibility, and repeatability (up to 7 times)	Dependent electron transfer rate on polymer thickness, optimization of illumination time, and monomer to cross-	[234]
AuNPs/ MXene@PAMAM	Covalent bonding	cTnT	0.1-0.001 ng/ mL	0.069 ng/mL	Enhanced loading capacity, conductivity, sensitivity, stability, large specific surface areas, in vitro biocompatibility	linker ratio Irreversible oxidation of MXene under anodic potentials, limited LOD	[235]
2D Black phosphorou Ca ²⁺ -doped BP@PDA	S _	Protein, DNA	P: 0.1–25 nM, D: 1.0–10 nM	P: 0.02 nM, D: 0.52 nM	Resistance to non-specific displacement by other biological ligands, simple preparation, highly stable, and biocompatible	Short biostability, selectivity, and sensitivity in complex biosamples	[236]
CuNPs-Chit-BP	Electrostatic attraction	H_2O_2	10 μM–10.3 mM	0.390 μΜ	Enhanced BP dispersion, increased surface area, electro-catalytic	Partial oxidation of BP over time	[237]
Poly-L-lysine-BP	Non-covalent electrostatic	H_2O_2	10 μΜ–700 μΜ	-	activity, and selectivity High direct electron transfer and electrocatalytic activity	Careful synthesis control process to maintain the native	[159]
BP/polyallylamine	attraction Electrostatic attraction	Interleukin-	0.003–75 ng/ mL	1 pg/mL	Economical fabrication, enhanced stability, charge transfer rate,	structure and activity of Hb Optimization of PAMI ratios for stable BP encapsulation	[160]

(continued on next page)

Table 1 (continued)

Biosensor designs	Interactions	Analytes	Linear range	LOD	Advantages	Challenges	References
Poly-L-lysine/BP- TFG	Electrostatic attractions	NSE	0.01–100 ng/ mL	1 pg/ mL ⁻¹	Label-free, ultrahigh sensitivity, specificity, real-time monitoring	Efficient layer-by-layer (LbL) technique, BP stable and uniform coating on fiber surface	[161]
Polyethyleneimine- BP	Electrostatic adsorption	Cu ²⁺	0.25–177 µМ	0.02 μΜ	Excellent electrochemical reduction, surface area, mechanical stability, flexibility, sensitivity (0.05–0.1 μ A μ M $^{-1}$ cm $^{-2}$) and selectivity, rapid response (1.5 s)	Short-term stability, optimization of PEI to BP ratio	[238]
2D Transition metal	dichalcogenides						
WS ₂ /B-PVA	-	HbA1c	$3.3\times10^{-8}~M$	-	Specific recognition of HbA1c achieved through reversible bonds with boronic acid moiety	Challenging exfoliation and functionalization of WS ₂ NSs in water	[171]
MoS ₂ /WS ₂ @PDA	Electron transfer	5 fC-DNA	0.005–200 nM	3.7 pM	Improved photoactivity, sensitivity, selectivity, and specificity, applicable for RNA formylation detection	Steric-hindrance effects and electrostatic repulsion caused by 5 fC-DNA	[172]
SPAN-MoS ₂	π – π * interaction	CAP	$0.11000~\mu\text{mol}$ L^{-1}	$\begin{array}{l} 0.65 \times \\ 10^{-8} \text{ mol} \\ L^{-1} \end{array}$	Enhanced electrocatalytic activity, reproducibility, long-term stability, and sensitivity	Optimization of mass ratio and ultrasonication time, pH sensitivity, alkaline interference	[239]
PANI-MoS ₂	Electrostatic attraction	DNA	$\begin{array}{c} 1.0 \times 10^{-15} - \\ 1.0 \times 10^{-6} \end{array}$	$\begin{array}{c} 2.0 \times \\ 10^{-16} \end{array}$	High electrochemical activity, efficient DNA immobilization, and hybridization	Optimization of MoS_2 dosage and reaction time	[240]
pDNA/PANI-MoS ₂ / ITO	Electrostatic attraction	CML	10^{-6} M $ 10^{-17}$ M	$3 \times 10^{-18} \mathrm{M}$	Large electroactive surface area, improved electron transfer, catalytic activity and conductivity, uniform growth of nanocomposite, high stability (up to 42 days)	Template removal, optimization of experimental conditions, specific detection	[241]
PIn6COOH/WS ₂	Physical adsorption	DNA	$\begin{array}{c} 1 \times 10^{-17} - 0.1 \\ \times 10^{-12} \ mol \\ L^{-1} \end{array}$	$\begin{array}{c} 0.23 \times \\ 10^{-19} \\ \text{mol L}^{-1} \end{array}$	Label-free, large specific surface area, high stability (up to 10 days), and selectivity.	Steric hindrance from ssDNA probe, specificity	[242]
WS ₂ -PEDOT/GO- SWCNTs/GCE	-	VB ₂	- 0.002–0.9 μM	0.7 nM	Improved response and electron transfer, specificity via imprinting, good anti-interference ability, and storage stability	Optimization of thickness, molar ratios, pH, and incubation time	[243]
2D Metal-organic fra	ameworks						
Co–Ni (Fe)-MOF/ PPy	-	Glucose	0.002–3.0 mM	1.13 μΜ	Superior electrochemical activity, sensitivity (1805 μ A mM ⁻¹ cm ⁻²), and selectivity	Low reproducibility and stability	[191]
MIP/Au@PANI/ SeS ₂ @Co MOF	-	PT	0.001–100 nM	0.66pM	Higher electrocatalytic activity, improved surface area, active sites, and sensitivity.	Electrochemical signal stability, steric hindrance due to PT imprinting	[244]
PFO- H ₂ dtoaCu	Adsorption	MMP-2	0.1–2.5 pg/mL	-	High specificity and selectivity, PFO dots as ultrasensitive luminophore, good stability	Optimization of reaction conditions	[245]
ZIF-L/PEDOT	Electrostatic attraction	DA	$25~nM-to~500\\ \mu M$	7 nM	High conductivity, catalytic activity, and sensitivity (381.3 μ A mM ⁻¹ cm ⁻²) at S/N = 3, long-term storage stability (up to 28 days)	Detachment of ZIF-L/PEDOT particles from electrode surface over time	[192]
Ab/Cu ₃ (BTC) ₂ - PANI/ITO	-	E. coli	$\begin{array}{l} 22\times10^8~\text{cfu/}\\ \text{mL} \end{array}$	$2.0~{ m cfu}$ ${ m mL}^{-1}$	Quick response (2 min), large surface area, high stability (up to 60 days), and selectivity	Biosensor stability during fabrication and use, specificity in complex sample matrices	[246]

EDC: 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide; NHS: N-hydroxy succinimide; CMF: Cellulose microfiber; ChOx: Cholesterol oxidase; EPNS: Electroconductive polymer nanosponge; GCE: Glassy carbon electrode; CIPs: Cell imprinted polymers; PAMAM: Poly(amidoamine); cTnT: Human cardiac troponin T; PDA: polydopamine; HbA1c: Glycated hemoglobin; SPAN: self-doped polyaniline; CAP: Chloramphenicol; ITO: Indium tin oxide; CML: Chronic myelogenous leukemia; VB₂: vitamin B₂; PT: Patulin; PIn6COOH: Poly(indole-6-carboxylic acid); PFO: Poly(9,9-dioctylfluorenyl-2,7-diyl).

bioimaging, photocatalysis, and more [79–81]. Furthermore, due to these characteristics, ultrathin 2D g- C_3N_4 materials have demonstrated minimal cytotoxicity and strong tissue compatibility, suggesting their suitability for applications in bioelectronics and biosensors [60,82]. They have effectively detected various substances, such as antibiotics, harmful metal ions, organic molecules, enzymes, nucleic acids, and chemical substances, leveraging their diverse electronic and semi-conductor features [83].

Despite its potential, g- C_3N_4 suffers from some limitations, like low conductivity [84], structural disorder [85], poor dispersibility [86], and limited processability [87]. Natural ligands like -N/-NH/-COOH and $-NH_2$ present on carboxyl-g- C_3N_4 exhibit strong chelate bonding capabilities with metal complexes or can be linked to biomolecules using

carbodiimide coupling chemistry [88]. This process entails chemical modifications like oxidation/reduction, amidation, and polymer grafting, improving stability at the expense of introducing defects in the material's $\rm sp^2$ framework that can impact electrical properties but can be advantageous in applications like catalysis and composites when the defect density is carefully controlled [85]. One promising approach to enhancing $\rm g\text{-}C_3N_4$ properties involves combining it with polymers via non-covalent and covalent modifications, offering unique functionalities such as biodegradability, stimulus responsiveness, self-healing, and electrical conductivity [89]. Combining traditional polymers with innovative metal-free sheets like $\rm g\text{-}C_3N_4$ holds significant potential. This fusion harnesses the strength of both material classes. $\rm g\text{-}C_3N_4$ contributes to improved photocatalytic and photoluminescent properties and

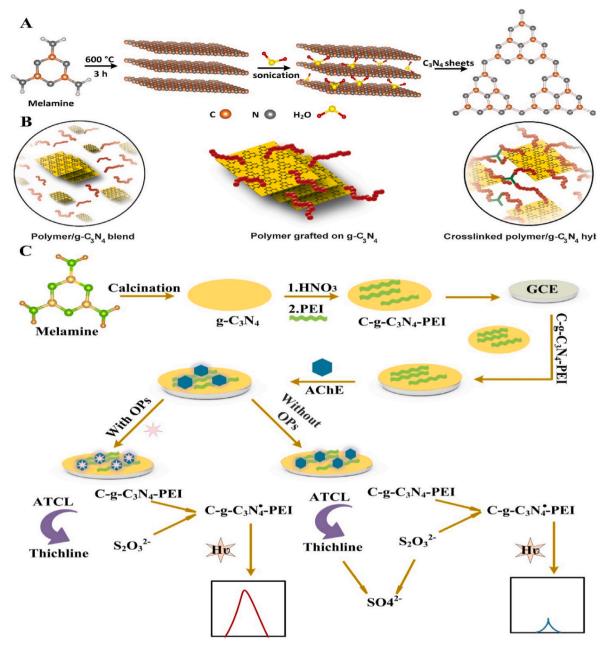


Fig. 4. A) Synthesis and structure of C_3N_4 (g- C_3N_4) sheets. Reproduced from Ref. [77]. with permission from *American Chemical Society*. **B)** Methods for g- C_3N_4 /polymer modification: techniques involving the incorporation of g- C_3N_4 into polymers through blending, grafting, and grafting/crosslinking processes. Reproduced from Ref. [87]. with permission from *Royal Society of Chemistry*. **C)** Illustrative depiction of the synthesis procedure for C-g- C_3N_4 -PEI and the construction of modified biosensor, including insights into the response mechanism. Reproduced from Ref. [94]. with permission from *Elsevier*.

improves mechanical and physical characteristics, whereas polymers enhance processability and conductivity [90]. Various techniques can be employed to create g-C₃N₄/polymer composite materials [87], including (i) "physical absorption or deposition" (typically in a liquid phase), (ii) the "grafting from" method involving polymerization from g-C₃N₄ surface, and (iii) the "grafting to" process that relies on chemical bonding between active polymer end groups and reactive sites on g-C₃N₄ (Fig. 4B).

ECL, also known as electrogenerated chemiluminescence, is a widely employed method for generating light from ECL emitters that are excited electrochemically through efficient electron transfer. ECL has gained prominence in biosensing due to its rapid response, high sensitivity, simplified setup, broad detection range, flexibility, cost-effectiveness, and compatibility with miniaturized instrumentation [91]. Since the revelation of $g\text{-}C_3N_4$ nanosheets' cathodic ECL characteristics, they have

become a focal point for ECL biosensors. While $g\text{-}C_3N_4$ has its advantages, researchers are exploring the integration of polymers to enhance the sensitivity and specificity of ECL sensors [87,92]. A signal-on ECL biosensor was fabricated by combining BSA and phenoxy dextran (DexP) with $g\text{-}C_3N_4$ for detecting concanavalin A [93]. The dexP- $g\text{-}C_3N_4$ nanocomposite was bound to Con A binding sites via a specific carbohydrate-Con A interaction, forming a sandwiched structure with GOx and Con A during electrode fabrication. Increased Con A concentration led to more DexP- $g\text{-}C_3N_4$ binding, enhancing the ECL signal for quantitative Con A detection. DexP- $g\text{-}C_3N_4$ was an ECL signal probe and facilitated specific Con A binding, enabling a novel sandwich-like ECL detection strategy. The biosensor showed remarkable sensitivity (LOD of 17 pg/mL for Con A) and a broad linear detection range (0.05–100 ng/mL), with stable performance over successive scans (RSD of 2.2 % for ten cycles), long-term storage (maintaining 91.5 % of the initial

response after 15 days), and reproducibility (RSD of 5.7 %). Additionally, it demonstrated practicality for detecting Con A in real biological samples, with recoveries ranging from 94.2 to 104 %. The critical contribution of this work was preparing the DexP- g-C₃N₄ hybrid and its application in constructing a signal-on sandwiched ECL biosensor. Although the biosensor had a longer preparation time, combining 3D-GR-AuNPs nanocomposites and GOx in advance can shorten the total assay time by avoiding prolonged GOx incubation on the electrode. In a different investigation, Wang and colleagues developed an ECL biosensor using a hybrid material, denoted as C-g-C₃N₄-PEI, along with acetylcholinesterase (AChE) for susceptible detection of organophosphate pesticides (OPs) [94]. This nanocomposite was synthesized by establishing a covalent amide connection between the amino unit of polyethyleneimine (PEI) and the carboxylic group of carboxylated-g-C₃N₄ using EDC/NHS coupling. In the absence of OPs, AChE catalyzed the hydrolysis of ATCl to produce thiocholine, consuming $S_2O_8^{2-}$ co-reactant and decreasing the ECL signal. In the presence of OPs, inhibition of AChE by OPs hindered thiocholine generation, leading to increased ECL response (as shown in Fig. 4C). The resulting functionalized biosensor for detecting OPs achieved an impressive LOD of 0.3 pM. Still, it was limited to standard OP solutions under optimal conditions, necessitating adaptation for real-life samples. Recovery experiments showed 97.4-102 % recoveries, affirming the biosensor's potential for sample analysis, while stability and reproducibility tests revealed an R.S.D. of 0.32 % and 2.36 %, respectively, indicating excellent stability and acceptable reproducibility.

The limited ability to disperse in organic and aqueous environments of pure g-C₃N₄ is one of the main challenges in utilizing it for practical applications, as solid van der Waal (π – π stacking) interactions cause accumulation [95]. However, there are two methods to enhance its dispersibility in the aqueous phase. One approach involves adding charged groups to promote stability through electrostatic repulsion in aqueous environments. Another method is to attach hydrophilic polymers, which use steric repulsion to stabilize g-C₃N₄ dispersion in water. As polymers have been the preferred choice for enhancing dispersibility, various methods exist for incorporating polymers. One straightforward approach to achieving dispersible g-C₃N₄ involved initiating polymerization using g-C₃N₄ through photoinitiation [96]. This involved adding N, N-dimethylacrylamide (DMA) to a dispersion of g-C₃N₄ in ethylene glycol/water and exposing it to visible light. This process led to the development of a gel with a loose structure, characterized by high viscosity and composed of PDMA-grafted g-C₃N₄, making it readily dispersible in water. In addition, these dispersions demonstrated excellent stability, lasting up to two months, thanks to the steric stabilization of g-C₃N₄ colloids. Furthermore, pH-responsive solubility can be achieved by using allylamine, resulting in solubility in acidic solutions and precipitation in alkaline conditions. A recent development involved grafting allylamine to g-C₃N₄ [97], notably enhancing its dispersion at an acidic pH of 4 through amino group protonation. Conversely, raising the pH to 9 resulted in rapid particle precipitation due to amino group deprotonation. Re-acidifying the solution restored uniform dispersion, indicating reversibility. In biosensor applications, the improved sensitivity and selectivity provided by polymer-functionalized g-C₃N₄ can be credited to improved dispersion and better accessibility to the sensing

Despite polymers offering new possibilities for g- C_3N_4 usage, there are still notable challenges to address. Limitations arise from the dependence on physical blending methods and employing environmentally unfriendly organic solvents in synthesis processes. To pave the way for large-scale production, future research should emphasize solvent-free and eco-friendly approaches [87]. Additionally, a comprehensive understanding of the electronic and lattice systems of polymer/g- C_3N_4 composites is crucial for unraveling their sensor mechanisms. Despite initial hurdles, advancements in polymer compatibility and dispersibility with g- C_3N_4 have been observed, marked by favourable interactions such as hydrogen bonding and π - π

interactions. Overall, g- C_3N_4 holds promise as an affordable and sustainable semiconductor. With careful adjustments to its properties and polymer design, it could significantly impact academia and various industries, particularly bioelectronic devices.

4.2. 2D graphene family

Since the groundbreaking isolation of high-quality graphene, this single-atom-thick layer of sp² hybridized C atoms has been spotlighted as the most extensively studied 2D material in material science [98,99] (Fig. 5A). 2D graphene, with its hexagonal patterns, boasts exceptional properties, including outstanding electrical conductivity, a high specific surface area, robust mechanical strength, rapid electron mobility, and ambipolar transfer abilities [100–102]. These distinctive features make graphene suitable for diverse applications, including bioelectronic devices and biosensors [103]. Moreover, its chemical surface modification has led to the creation of reduced graphene oxide (rGO) and graphene oxide (GO) as cutting-edge innovative materials [104]. Being hydrophilic, GO possesses numerous oxygenated functional groups like -OH, -C=O, and -COOH, enabling it to capture water molecules from the surroundings, making it an ideal biosensing nanomaterial due to its affinity towards aromatic rings and hydrophobic bio-molecules, good water dispersibility, and fluorescence properties [105,106].

Oxygenated groups on both the basal and edge planes of rGO create defect sites, enhancing its electrocatalytic activity more than graphene. This increased activity is observed towards various electroactive species, including H₂O₂, glucose, etc. [107]. Hence, graphene and its derivatives have made strides in applications like biosensors and bioelectronics due to their exceptional properties like high absorptivity, superior conductivity, and stability, making them excellent materials for electrodes with swift carrier mobility [108,109]. For instance, a biosensor was innovatively crafted using GO as an efficient fluorescent marker and Au NPs as a fluorescence quencher for detecting DNA-DNA interactions. This study highlighted the potential of large-scale, cost-effective GO nanomaterials in various bio-applications, such as biosensors and molecular imaging [110]. However, the LOD of approximately 0.1 pmol of target DNA limits its applicability in scenarios requiring detection at lower concentrations. Moreover, the experimental conditions and protocols do not fully reflect real-world complexities, potentially affecting the biosensor's performance in practical applications.

Hence, in spite of their outstanding physiochemical and biological advantages, that regard them as ideal materials for developing biosensors, graphene-based nanomaterials suffer from some limitations. This includes aggregation due to weak van der Waals attractions, limited biorecognition capabilities, and varying levels of biotoxicity, particularly in the case of GO, where the extent of toxicity is contingent on concentration and exposure time [43,111,112]. To address these issues, enhancing graphene and its derivatives through biological or chemical functionalization is crucial to improving recognition and biocompatibility. This improvement is possible because functional moieties, as seen in rGO and GO, permit surface functionalization with appropriate components, facilitating the attachment of biomolecules, metal oxides, and polymers to enhance biosensors' performance [6].

Various strategies, categorized as non-covalent and covalent functionalization, are employed to functionalize graphene nanomaterials with substances like polymers. These strategies resemble those used for other 2D materials [43,113]. In the non-covalent method, negative rGO/GO (carry negative charges due to their oxygenated groups) or functionalizing polymers like PEI, PEG, chitosan, PANI, etc., can positively charge-neutral graphene, facilitating electrostatic cross-linking with biomolecules carrying a negative charge, For instance, a susceptible DNA sensor for detecting the HIV-1 gene was created using a GCE modified by graphene-Nafion nanocomposite film through $\pi-\pi^*$ stacking interactions [114]. The biosensor operated by adsorbing ssDNA onto the modified surface of a GCE, which then interacted with the target DNA to form dsDNA. This interaction induced changes in electron transfer

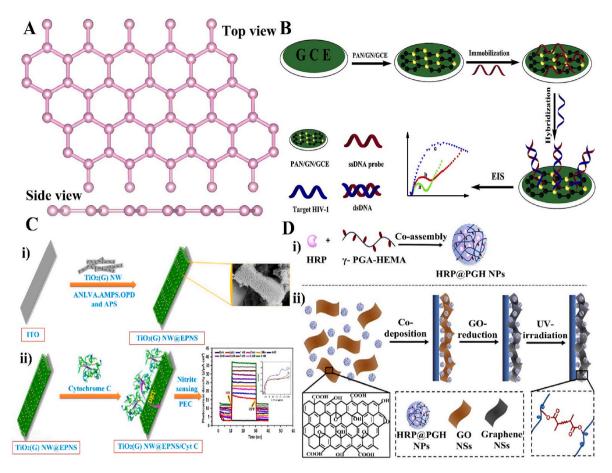


Fig. 5. A) Structure of 2D graphene (top and side view). B) Illustration of the detection process for the HIV-1 gene, accompanied by the sequential synthesis steps and the performance evaluation of the DNA sensor. Reprinted from Ref. [111]. with permission from *Elsevier*. C) i) Synthesis of EPNS coated with $TiO_2(G)$. ii) Immobilization of Cyt C to create a biosensor using $TiO_2(G)$ NW@EPNS, followed by photoamperometric biosensing. Reproduced from Ref. [117]. with permission from *Elsevier*. D) Schematic representation of enzymatic biosensor, i) Formation of enzyme-polymer nanoparticles (HRP@PGH NPs) through the co-assembly of HRP and γ-PGA-HEMA. ii) Construction of integrated HRP@PGH/GNS sensing film by electrophoretic co-deposition of 0D HRP@PGH NPs and 2D GO NSs. Reproduced from Ref. [120]. with permission from *Elsevier*.

resistance, allowing for the detection of the HIV-1 gene. Notably, Nafion stabilized graphene and enhanced its dispersion, enabling sensitive detection of a target gene over a wide concentration range with a LOD of 0.23×10^{-15} M. The biosensor also exhibited good selectivity, stability, and reproducibility, making it a promising tool for HIV-1 gene detection. However, further discussion on biosensor specificity in actual samples, particularly in distinguishing between complementary and mismatched DNA sequences, would strengthen the evaluation of its performance.

Considering the robust stability associated with covalent bonding, methods involving covalent modification play a key role in functionalizing graphene-based nanomaterials with polymers [115]. Oxygenated functional moieties like hydroxyl and carboxyl groups on rGO and GO surfaces provide optimal platforms for immobilizing polymers/biopolymers with multiple functional groups [108]. The -COOH moieties in GO and rGO are particularly advantageous for achieving higher biomolecule loading through robust amide bond formation with aminated groups present in polymers like PEI, PANI, and biopolymers such as proteins, enzymes, antibodies, and peptides, assisted by EDC and NHS interactions [116]. In 2019, Xiang and coworkers addressed the agglomeration issue of graphene materials by developing a novel DNA using ssDNA/PANI/graphene nanocomposites through reverse-phase polymerization [111]. This biosensor was designed for detecting HIV-1 and featured a covalently immobilized ssDNA probe on a PAN/GN/GCE (Fig. 5B). Results indicated that PANI introduction not only improved graphene dispersion in N, N-dimethylformamide (DMF) but also enhanced the stability and conductivity of the modified

electrode for HIV-1. The DNA biosensor demonstrated high sensitivity to HIV-1 gene concentrations (5.0×10^{-16} M to 1.0×10^{-10} M) with an LOD of 1.0×10^{-16} M (S/N = 3), attributed to graphene and PAN synergy. In addition, it exhibited selective hybridization with complementary DNA sequences, showing stable reproducibility (standard deviation of 4.7 %) and storage stability for two weeks at 2 °C. The study suggests its clinical utility for HIV-1 diagnosis. But it overlooks key practical challenges like sample handling, assay time, and cost-effectiveness, which are pivotal for real-world deployment.

An innovative PEC biosensing platform was created to improve the biocompatibility and photoactivity of 2D graphene. The electroconductive polymer nanosponge (EPNS), containing graphene (G) and TiO₂ nanowires (NW) with cytochrome C (Cyt C) immobilization, served as the foundation for constructing a highly selective and sensitive PEC biosensor for detecting NO_2^- [117]. The proposed biosensing mechanism involved light-induced electron transfer from TiO₂ to G that facilitated the transformation of heme-Fe (III) to heme-Fe (II) in immobilized Cyt C, where NO interacted with heme-Fe (II) sites to form N2O, elucidating the role of NO rather than NO2 in heme-protein interactions. The biosensor enhanced performance at low bias potential (-0.11 V). It offered advantages such as sound sensitivity (201.79 μA μM⁻¹ cm⁻²), rapid response (~5s), a wide linear range from 0.5 to 9000 mM, stability, and selectivity for detecting NO₂ in actual samples (Fig. 5C). The fabrication process involved multiple steps in this study, which may hinder scalability and reproducibility. Additionally, environmental factors like light intensity and pH can affect the sensitivity and reliability of PEC

biosensors. Moreover, graphene-based materials are employed to detect microorganisms. A gold electrode modified using thin active layers of rGO/PEI was developed through electrophoretic deposition (EPD) for detecting uropathogenic E. coli [118]. The electrochemical analysis showed improved redox current upon rGO/PEI electrode functionalization, suggesting enhanced electron transfer kinetics. However, the electrode's specificity towards E. coli UTI89 was limited, especially at lower concentrations, posing challenges for accurate detection, particularly in complex samples. Anti-fimbrial E. coli antibodies were immobilized on the electrodes using NHS/EDC coupling chemistry to improve specificity. The sensor exhibited a linear range from 10^1 – 10^4 cfu mL⁻¹ and an LOD below 10 cfu mL⁻¹ for UTI89 detection with a promising sensitivity. Notably, the biosensor also performed well in various media, including aqueous, urine, and serum, demonstrating its potential for diagnosing pathogenic viruses. However, exploring additional strategies, such as optimizing antibody immobilization density, may further enhance specificity.

The integration of polymers with graphene is also a keen area of study because it enhances the sensitivity of biosensors due to their unique structural and compositional sophistication [119]. A robust enzymatic biosensing platform was fabricated by depositing OD enzyme-loaded biopolymeric NPs onto 2D GO NSs, which acted as conductive nano-supports (Fig. 5D). This platform demonstrated outstanding performance for H₂O₂ due to a synergistic effect from material design and the hierarchical OD-2D nanostructure [120]. The photo-cross-linkable polypeptide γ-PGA-HEMA played a pivotal role in developing a biocompatible microenvironment and enhancing the bioactivity and structural stability of the HRP enzyme through photo-induced cross-linking. Concurrently, the conductive graphene NSs facilitated electron flow across the sensing interface, elevating sensitivity. The HRP@PGH/GNSS sensor demonstrated rapid H2O2 detection within 3s, with a linear range of 1.0-340 μM and a LOD of $0.27~\mu\text{M},$ offering enhanced sensitivity and stability. It exhibited high anti-interference ability and excellent reproducibility, making it suitable for practical applications in biomedical fields with recoveries ranging from 96.80 % to 104.46 % in human serum samples. However, the study faces challenges in ensuring long-term stability and preventing enzyme leakage due to the complex synthesis process and potential interactions between components.

Although the potential of polymer-functionalized graphene for biosensors is promising, several hurdles persist. Challenges lie in effectively applying 2D graphene-based biosensors, especially for real sample analysis, which necessitates the creation of highly selective sensors that excel in complex conditions [58]. Moreover, non-covalent functionalization methods may lead to false positive signals due to the adsorption capacity of graphene for non-target substances. Conversely, covalent strategies, while effective, are complex and require precise control. Last but not least, ensuring the uniformity of these materials in terms of thickness, size, the number of surface functional moieties, and biocompatibility is a significant hurdle [43]. This uniformity not only impacts the efficiency of surface functionalization but also has a profound impact on the reproducibility and performance assessment of the developed biosensors. There is an optimistic anticipation that ongoing research efforts will overcome these issues, leading to a more sophisticated graphene biosensor system.

4.3. 2D MXenes nanostructures

MXenes represent an innovative class of 2D materials characterized by layered structures of metallic nitrides, carbides, or carbonitrides. The formation of MXenes involves selectively removing "A" layers from MAX phases with the general formula " $M_{n+1}AX_n$ ", where "M" signifies transitional metals like V, Ti, Mo, "A" signifies main-group elements from IIIA or IVA, "X" signifies carbonitrides, nitrogen, or carbon, and "n" represents the number of layers linked by van der Waals forces and does not exceed 3. MAX phases exhibit hexagonal layering, where X atoms

occupy the octahedral positions within tightly packed M layers, with A atoms interspersed between neighbouring layers (Fig. 6A). During synthesis, "M" and "X" remain unchanged from their MAX predecessors. At the same time, "A" groups are introduced during the etching stage to signify terminated groups like chlorine (–Cl), fluorine (–F), hydroxyl (–OH), and oxygen (–O) [121,122], as illustrated in Fig. 6B. This process makes them hydrophilic, near-infrared (NIR) absorption capable, metallic conductive, size-adjustable, biocompatible, and opens up opportunities for surface modifications [42,123]. However, the etching process significantly impacts their suitability for biomedical purposes due to a lack of controlled drug release, instability in physiological environments, and the potential harm to healthy cells or tissues due to limited target specificity [42]. Even tiny illustrations of remnants of HF in biological devices can trigger cell death within living organisms [124].

To overcome these limitations, researchers have enhanced MXenes by introducing surface modifications. This process involves incorporating various materials such as polymers like PVP [125], polyvinyl alcohol (PVA) [126], cellulose [127], soybean phospholipid (SP) [128], PEG [129], or inorganic nanoparticles like mesoporous silica nanoparticles (MSNs) [130] and polyoxometalates (POMs) [131]. Biocompatible polymers are particularly favoured due to their non-toxic nature, compatibility with biological fluids, and the absence of a need for surgical removal after treatment [132]. Surface modifications using polymers like SP, PEG, PVA, PVP, and PLGA enhance the physical and chemical stability of MXenes and make them biocompatible. These modifications can be achieved through surface adsorption or electrostatic interactions [42]. Furthermore, MXenes can detect different analytes by altering their conductivity through charge transfer while attaching various molecules. This attachment makes MXenes highly adaptable for sensing applications with varying surface terminations. Rakhi and his colleagues developed an innovative biosensor called GOx/Au/Ti₃C₂/Nafion/GCE, designed for glucose detection [133]. In this system, the enzyme glucose oxidase (GOx) was crucial in reducing glucose to gluconolactone and H2O2. The high reduction/oxidation potential of H₂O₂ led to GOx immobilization on MXene/GCE for glucose sensing. Including Nafion, improved enzyme adhesion to GCE, elimination of interfering signals, and enhanced biosensor selectivity.

To maintain enzyme activity and promote direct electron transfer, gold nanoparticles (Au) were employed during adsorption, reducing insulating effects and increasing electrical conductivity between GOx and GCE. MXenes, with their expansive surface area, facilitated charge accumulation, thereby boosting sensor sensitivity and linearity. The GOx comprised one flavin adenine dinucleotide (FAD) molecule and two protein units, explaining the glucose-sensing mechanism (Eqs. (1)–(3)),

$$GOx (FAD) + Glucose \rightarrow GOx (FAD_2) + Glucono - d - lactone$$
 (1)

$$GOx (FADH2) + O2 \rightarrow GOx (FAD) = H2O2$$
 (2)

Glucose
$$+ O_2 \xrightarrow{GOx}$$
 Glucono $- d$ – lactone $+ H_2O_2$ (3)

This MXene/GCE biosensor exhibited outstanding storage stability, a good sensitivity (4.2 μ A mM $^{-1}$ cm $^{-2}$), a wide detection range (0.1–18 mM), and a low LOD (5.9 μ M). Yet, evaluating the biosensor's performance with real biological samples, such as blood or urine, would validate its practical utility and reliability in clinical or point-of-care testing (POCT). Recently, a cutting-edge biosensing patch designed for personal wearable bioelectronics was introduced. This advanced patch is constructed using a unique combination of materials, including MXene/fluoropolymer nanofiber-derived hierarchical porous TiO₂ and 3D fibrous carbon nanohybrid electrodes, which resulted in numerous active edges to facilitate efficient electron transmission, remarkable electrical conductivity (with a sheet resistance of 15.6 Ω sq $^{-1}$), and ample surface area for enzyme immobilization [134]. Three complementary effects contribute to these exceptional attributes: first, the

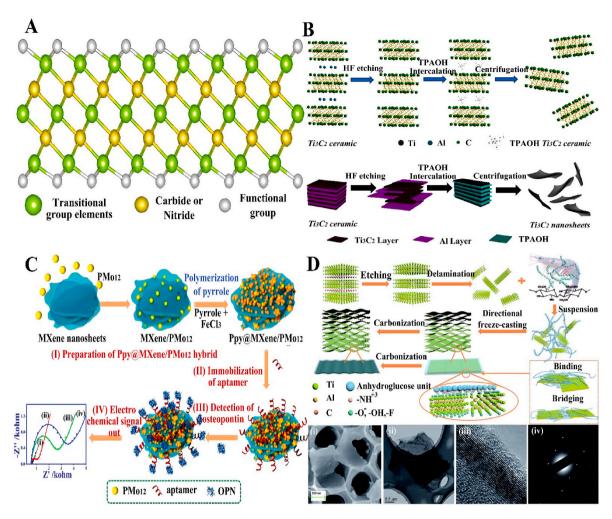


Fig. 6. A) Structure of 2D MXene. **B)** Diagram depicting the two-step exfoliation process for producing Ti₃C₂ nanosheets. Reproduced from Ref. [128]. with permission from *Wiley*. **C)** Schematic illustration detailing the assembly of the aptasensor for OPN detection via PPy@Ti₃C₂T_x/PMO₁₂ hybrid. The process involves **i)** the synthesis of the PPy@Ti₃C₂T_x/PMO₁₂ hybrid, **ii)** immobilization of the aptamer, **iii)** OPN detection, and **iv)** the subsequent output of the electrochemical signal. Reprinted from Ref. [136]. with permission from *Elsevier*. **D)** Process depiction for creating MXene and CS_x/MXene-C aerogels. **i)** SEM depiction of MXene sheets on the AAO template. **ii)** TEM image showcasing the MXene sheet. **iii)** Close-up view of the MXene sheet in TEM. **iv)** Patterns of SAED for the MXene. Reprinted from Ref. [137]. with permission from *Royal Society of Chemistry*.

conductivity of the inner, unoxidized MXene layers; second, the rapid electron transfer facilitated by the exterior TiO2 NPs; and third, electron "bridge" effects provided by the porous disordered carbon structure. When incorporated into textiles, this modified biosensing patch precisely monitors sweat glucose levels while adjusting pH (exhibiting a sensitivity of 77.12 µA mm⁻¹ cm⁻² within physiological levels ranging from 0.01 to 0.2×10^{-4} m and recording electrocardiogram signals (with a S/N of 37.63 dB). Overall, this study represents a significant advancement in wearable biosensing technology. However, further research is needed to address scalability, long-term stability, and compatibility with diverse physiological conditions. Moreover, Wang and his colleagues developed a biosensor for H2O2 detection by anchoring hemoglobin (Hb) onto TiO2-modified Ti3C2 [135]. This enhances conductivity and surface area by promoting direct electron transport between the electrode and protein. Nafion usage eliminated interference, enhanced selectivity, and improved enzyme adhesion to the electrode. Adding TiO₂ nanoparticles enhanced protein stability and bioactivity, improving biosensor performance. They proposed an electrocatalytic mechanism for the Hb-based electrode, as mentioned in Eqs. (4) and (5).

$$Hb (Fe^{+3}) + e^{-} \rightarrow Hb (Fe^{+2})$$
 (4)

Hb
$$(Fe^{+2}) + H_2O_2 + 2H^+ \rightarrow Hb (Fe^{+3}) + 2H_2O$$
 (5)

This biosensor exhibited rapid response time (<3 s), with a broad detection range of 0.1–380 mM, a sensitivity of 447.3 μ A mM⁻¹ cm⁻², and a LOD of 14 mM for H₂O₂ detection. Additionally, it showed excellent stability (97.5 % retention after 30 days) and reproducibility (R.S.D of 2.1 %) for practical applications. However, achieving uniform TiO₂ nanoparticle distribution on Ti₃C₂ layers is crucial for consistent biosensor performance; variations may lead to heterogeneous enzyme immobilization, impacting sensitivity and reproducibility. In 2019, a PPy@Ti₃C₂T_x/PMo₁₂-based electrochemical aptasensor was devised to detect the cancer biomarker OPN (osteopontin) by employing a combination of phosphomolybdic acid (PMo₁₂) nanoparticles and Ti₃C₂T_x NSs within a polypyrrole (PPy) matrix (Fig. 6C). This aptasensor demonstrated remarkable sensitivity to OPN, achieving an impressively low LOD (0.98 fg mL⁻¹) and a more comprehensive detection range (0.05–10.0 ng/mL) [136]. The exceptional detection efficiency of this aptasensor can be credited to several critical factors, including the unique structure of the PPy layer with a π - π conjugated electron system, which enhances both electrochemical activity and OPN aptamer immobilization. Additionally, the aptasensor benefited from a π - π conjugated carbon nanostructure, good electron transport capabilities, and abundant TiO2 and Ti3C2Tx, preventing dense PPy stacking and aiding aptamer anchoring. Lastly, the strong binding affinity between polyoxometalates and biomacromolecules was primarily due to electrostatic interactions. Real sample analysis using human serum spiked with OPN confirmed the applicability of the aptasensor for OPN detection, with recoveries ranging from 96.61 % to 104.09 %. A discussion on the practical implications and potential challenges of deploying the aptasensor in real-world clinical settings would be valuable.

MXenes, with their adaptable hydrophilicity and surface characteristics, also hold significant promise for use in biomedical contexts. An example of this potential is evident in the nanosheets of ${\rm Ti}_3{\rm C}_2$ integrated into chitosan-based aerogels (depicted in Fig. 6D). These nanosheets exhibited impressive qualities, including flexibility, remarkable compressibility (up to 150,000 cycles), structural stability, and

elasticity, enabling them to endure extreme strain (up to 99 %), and repetitive bending [137]. Also, the unique lamellar structure of aerogel boasted an extensive linear range (0.5–70 % strain) and an extraordinarily high sensitivity (measuring 80.4 kPa⁻¹). These advantageous features position the aerogel as a promising material for flexible wearable devices that detect biological signals. However, the study does not extensively discuss the potential impact of environmental factors, like humidity or temperature, on the performance of the aerogels, which could be relevant for real-world applications.

Several significant challenges hinder the practical utility of utilizing polymer-functionalized 2D MXenes in bioelectronics. These obstacles include issues related to chemical stability (such as degradation in humid or high-temperature environments, particularly during storage

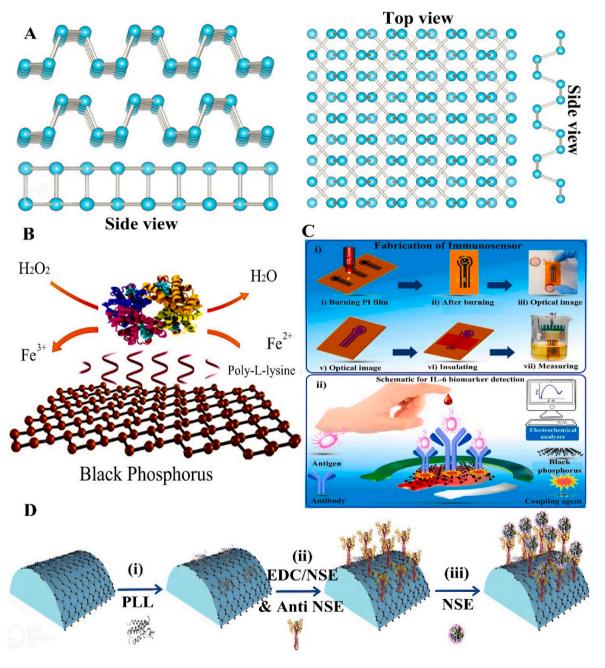


Fig. 7. A) Structure of 2D BP with top and side views. B) Poly-L-lysine/BP hybrid with detection mechanism. Reprinted from Ref. [159]. with permission from American Chemical Society. C) i) Steps involved in creating the IL-6 immunosensor, and ii) Diagram illustrating the detection process of the IL-6 biomarker. Reprinted from Ref. [160]. with permission from Elsevier. D) Functionalizing and linking BP-TFG: i) Incorporation of biocompatible PLL for biofunctionalization, ii) Fixation of anti-NSE through EDC/NHS bioconjugation, iii) Establishment of bio-affinity binding between anti-NSE and target NSE biomarkers. Reprinted from Ref. [161]. with permission from Elsevier.

and use), precise composition control, in vivo retention, mass production, and long-term biosafety [138,139]. Current methods, like hydrophilic polymers, enhance the water dispersity of MXenes in biosensors, but stability issues arise in extreme conditions. Future efforts should prioritize stable covalent modifications and dynamic bonding methods for biosensors. Additionally, research on functionalized 2D MXene biocompatibility remains nascent, particularly concerning cell uptake and toxicity [140]. Addressing these gaps requires developing novel surface modification techniques for biosensors and assessing the toxicity of polymer-functionalized 2D MXenes. The utilization of MXene-polymer combinations in biotechnology is in its early stages, and investigating polymer-based MXenes for bioelectronic applications is essential.

4.4. 2D black phosphorus

BP, also known as phosphorene in its monolayer form, has garnered significant interest since being rediscovered as a single-element 2D material in 2014, when liquid-phase-exfoliated BP demonstrated its effectiveness as a FET [141,142]. Unlike graphene, BP has a distinct puckered honeycomb orthorhombic lattice structure where each phosphorous atom covalently bonds with three neighbouring P atoms, resulting in a layer-to-layer spacing of about 5 Å [143,144], as depicted in Fig. 7A. This unique structure imparts BP with remarkable electronic characteristics, including high mobility (~1000 cm²Vs⁻¹), layer-dependent fluorescence, an adjustable direct band gap (bulk, 0.3 eV; monolayer, 2 eV), anisotropic electrical and thermal conductivities, and excellent optical response [55,145]. These distinctive electronic and optoelectronic properties position phosphorene as an ideal candidate for diverse applications, like optoelectronic devices, bioelectronic devices, solar cells, biosensors, and thermoelectric devices [146,147]. Also, the biocompatibility and low toxicity of 2D BP materials make it an attractive choice for biosensor and bioelectronic applications [148]. Its biocompatibility stems from its essential role in the human body, where phosphorus and phosphate constitute roughly 20 % of bones and teeth [149].

Additionally, by manipulating the bandgap of BP, its photodetection range can be adjusted when employed in fluorescence biosensors as photosensitizers. The 2D structure of BP allows for compact photodetectors in fluorescence biosensors, facilitating the development of portable biosensors [149]. These approaches may also apply to emerging 2D materials in the future. For the first time in 2017, Teng Yew and his team used 2D BP as a nanoscale fluorescent marker to fabricate a sensor for detecting DNA. This method relied on the varying affinities of labeled ssDNA and dsDNA fragments to self-assemble with BP nanoparticles [150]. The resulting nano-biosensor had a detection range of 4.0-4000 pM, showing high linearity (r = 0.91) and the ability to differentiate triple nucleotide differences. Additionally, it exhibited good sensitivity, with a LOD of 5.9 pM. The researchers suggested that 2D BP holds substantial promise for diverse biosensing applications, particularly in DNA nano biosensors, with potential applications for enzymes, proteins, and inorganic fluorescence-based methods. However, the study involved the following limitations: the complex synthesis process of BPNPs and the need for further assessment of selectivity, sensitivity, biocompatibility, and compatibility with real sample matrices.

Despite the promising research conducted on pristine 2D BP for applications such as photodetection, biosensors, and biomedicine, further progress is hindered by the inherent properties of pure BP [151,152]. Firstly, few-layer BP are vulnerable to oxidation in water and oxygen presence, which limits their practical applicability. This oxidation occurs because of the lone-electron pairs carried by surface P atoms [153]. Secondly, there is currently no effective method for preparing large-area single-crystal BP. Current top-down procedures like mechanical milling and liquid-phase exfoliation result in minor areas or create diminutive BP flakes. At the same time, bottom-up methods need high pressure and

heat to convert red phosphorus into BP [154–157]. Lastly, the poor solubility of BP in common organic solvents complicates its solution-based fabrication [158]. To address these limitations and enable diverse applications of BP, it is crucial to explore modification methods that can transform its unfavourable properties into desirable ones

Considering the progress made by materials like 2D graphene and MXenes, combining 2D BP with biocompatible materials seems promising to create a more stable and sensitive biosensing platform [159]. Scientists have demonstrated that modifying BP with polymers maintains its transport properties and enhances stability, making it more suitable for ambient conditions [55]. Polymer surface modifications can also improve BP chemical properties, including biocompatibility, and address issues like solubility in organic solvents [149]. Two main strategies for polymer-based surface modifications to enhance BP stability are covalent functionalization, which forms covalent bonds between polymer functional groups and BP surface lone-electron pairs. Noncovalent functionalization relies on interactions like hydrophobic and electrostatic forces between the two materials [15]. For example, a straightforward method involving non-covalent modifications was suggested to produce a hybrid material called pLL-BP through water-based exfoliation in aqueous conditions [159]. In this process, pLL is attached to the BP surface via hydrophobic interactions with butyl chains and electrostatic interactions with protonated amino groups on pLL and deprotonated phosphate (P_xO_y) groups on 2D BP (Fig. 7B). This resulting pLL-BP hybrid served as an excellent framework for immobilizing hemoglobin. It facilitated direct electron transport due to BP's remarkable conductivity and biological compatibility, preserving hemoglobin's natural structure and bioactivity. The enzymatic electrochemical biosensor, incorporating Hb-pLL-BP, demonstrated impressive catalytic efficiency in reducing oxygen and H₂O₂. Its electrochemical response to the concentration of H_2O_2 was within the 10 μ M–700 μ M range, making it a promising component for new biosensors and bioelectronics. While the study focused on the electrochemical performance of biosensors, it did not extensively study the analytical parameters like selectivity, stability, reproducibility, etc. Understanding these parameters could provide valuable insights into the biosensor's performance and stability in real sample analysis.

A different study developed an immunosensor for detecting the IL-6 biomarker by incorporating BP with a versatile cationic polymer called polyallylamine (PAMI) through electrostatic interaction [160]. This study was the first to address the BP degradation issue, improve charge transfer, and introduce –NH₂ groups for attaching target biomolecules. It was noted that the cationic properties of PAMI enhanced the stability of immobilized biomolecules by creating a saline environment (Fig. 7C). Also, the PAMI-coated BP platform provided abundant binding sites, resulting in exceptional stability, cost-effective production, and the potential for point-of-care (POC) analysis of cancer. The developed immunosensor showed linear responses for detecting IL-6 biomarkers from 0.003 to 75 ng/mL with a 1 pg/mL LOD, demonstrating high selectivity and reproducibility, with a recovery range of 93.1-98.4 % in spiked serum samples. Significantly, the researchers believed that BP-PAMI could be employed to recognize various disease biomarkers following treatment with respective biomolecules. Still, clinical validation using a larger sample size and comparison with established methods is necessary to evaluate its accuracy and reliability in real-world clinical settings. Apart from depositing functionalized BP onto an electrode, an optical biosensor using a three-step in-situ layering process is devised to apply BP onto a tilted fibre grating (TFG) to diagnose cancer [161]. Initially, they functionalized -OH groups on the optical fibre through an alkaline treatment. Subsequently, -OH groups on the fibre interacted with APTES, forming Si-O-Si bonds that created a positively charged surface. Finally, a few layers of BP attached to the fibre via the coulombic attraction between the positively charged APTES bonds and the BP surface (as shown in Fig. 7D). To facilitate cancer diagnosis, poly-L-lysine was integrated with the BP-TFG and acted as a connector

to connect with anti-NSE to target NSE biomarkers. The BP-TFG biosensor demonstrated an excellent linear response (0.01–100 ng $\rm mL^{-1})$ with a 1.0 $\rm pg/mL^{-1}$ LOD. The expressed sensitivity is 100 times more significant than GO and AuNP-based biosensors. However, the long-term stability and durability of the BP nanosheets deposited on fibre optic devices must be investigated. Environmental conditions, mechanical stress, and chemical exposure may affect the integrity and performance of the biosensor over time.

Although the promise of BP modification is significant, this research area is still in its early stages. Polymer-functionalized BP has promising applications in biosensors, enabling the attachment of biomarkers, but challenges remain [55]. Firstly, the preparation methods for BP/polymers need refinement, focusing on simplifying processes and achieving precise component additions in conventional single-crystalline methods. This characteristic becomes particularly crucial in disease diagnostics, such as tumor detection. Secondly, the judicious selection and

utilization of polymers, encompassing factors like polymer type, functionalization procedures, and polymer thickness, are vital for enhancing BP/polymer stability. Adjustable stability is valuable for practical applications, especially in electronics. Lastly, comprehensive research on the biocompatibility of BP/polymers, particularly their effects on the nervous, reproductive, and immune systems, is essential for their application in various fields, especially biomedicine.

4.5. 2D transition metal dichalcogenides

Layered TMDCs, considered one of their foremost graphene counterparts, have attracted substantial attention in recent decades due to their chemical adaptability and outstanding physicochemical attributes, including impressive mechanical features, good optical absorption efficiency, expansive surface areas, noteworthy electronic conductivity, robust chemical stability, high catalytic activities, an energy band gap

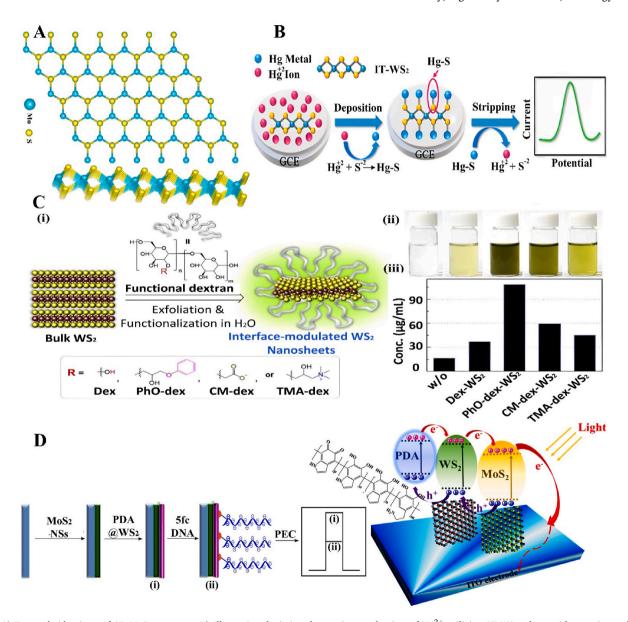


Fig. 8. A) Top and side views of 2D MoS_2 structure. B) Illustration depicting the sensing mechanism of Hg^{2+} utilizing 1T-WS₂, along with experimental results highlighting the S-Hg interaction. Reproduced from Ref. [168]. with permission from *Wiley*. C) Liquid-phase exfoliation of WS₂ nanosheets using various functionalized polymers (dextran, phenoxy-modified dextran, carboxymethyl dextran, and tri-methyl ammonium-modified dextran) in water. i) Schematic illustration of the process. ii) Photographs showcasing the exfoliated WS₂ nanosheets. iii) Concentrations of the nanosheets with different polymeric interfaces. Reproduced from Ref. [170]. with permission from *Elsevier*. D) Diagram depicting the PEC biosensor platform designed to detect 5 fC. Reproduced from Ref. [172]. with permission from *Elsevier*.

(0.2-2.1 eV) and straightforward synthesis methods [162,163]. These qualities hold promising potential for advancing high-performance biosensing applications [164,165]. TMDCs are layered materials composed of stacked planar crystals denoted by the general formula MX₂; here, M represents the transitional metallic atom, and X signifies the neighbouring chalcogen atoms, e.g., S, Mo, Se [166]. This structure, depicted in Fig. 8A, involves covalent bonding within the hexagonal lattice of the same layer and weak van der Waals forces between adjacent layers, facilitating easy thinning of the stack [167]. Examples of extensively researched TMDCs include WSe2, WS2, MoS2, and MoSe2. The electronic characteristics of these materials vary based on the elements and structural arrangements, providing flexibility for diverse applications [166]. Rahman and coworkers developed a straightforward hydrothermal method to prepare metallic 1T-WS2 microflowers for detecting trace ${\rm Hg}^{2+}$ levels [168]. The diagram illustrates the sensing process, where ${\rm Hg}^{2+}$ ions form strong adsorption and deposition on the 1T-WS₂ surface, creating a Hg-S covalent bond (Fig. 8B). The oxidization of mercury during stripping generated a current response proportional to the levels of ${\rm Hg}^{2+}$ in solution. The modified GCE exhibited a broad linear response to Hg^{2+} within the range of 1–1 μM and 0.1–1 mM, with an impressive LOD as low as 79.8 pM. While the 1T-WS₂ performance in tap water analysis showed promising results, its stability in real environmental conditions, such as varying pH levels, temperatures, and exposure to different substances, is not thoroughly investigated.

As discussed earlier, atomically thin TMDCs exhibit various advantageous properties like high photothermal response, fluorescence, and moderate conductivity, making them valuable for biosensor applications. However, these materials lack intrinsic biocompatibility and necessitate additional functionalization for integration into biological systems [164]. Moreover, further modification may be essential to improve sensitivity because most 2D TMDCs struggle to maintain a stable suspension in typical polar solvents like H2O, posing challenges for in vivo applications [164,169]. Due to their significant surface-to-volume ratio, one approach involves non-covalently incorporating biocompatible polymers into TMDC materials to create polymer/TMDC composites, facilitating their use in biosensing [164]. For example, a PEC sensing platform for detecting glutathione (GSH) was developed using PANI and nanoMoS2 composites as optoelectronic materials [61]. The PEC response increased with an increasing dose of PANI/nanoMoS₂, up to a certain point beyond which it decreased due to excessive weight hindering light transmission. The pH of the solution was also found to affect PEC intensity, with pH 6.0 identified as the ideal pH for subsequent studies. At this pH, the biosensor demonstrated a linear response (1.0×10^{-10} – 1.0×10^{-4} mol L⁻¹) for GSH concentrations, achieving a LOD of 0.31×10^{-12} mol L⁻¹. It displayed long-term stability, retaining over 90 % of its initial photocurrents after four weeks of storage. The method proved reliable for detecting intracellular GSH levels in human cell lines and GSH concentrations in human serum samples, with recoveries between 95.9 % and 103.3 %. The biosensor's performance was optimized at pH 6.0, which may not be suitable for all sample types or clinical scenarios, potentially impacting sensitivity and accuracy in samples with varying pH levels. In another study, dextran-polymer-functionalized WS2 NSs were created through liquid-phase exfoliation, resulting in four distinct WS2 interfaces [170]. These interface variations significantly influenced the thermodynamics and adsorption kinetics of a fluorescein-labeled DNA (FAM-DNA) probe. Compared to pristine dextran-WS2, (tetramethylammonium-modified dextran) TMA-dex-WS2 and (carboxymethyl-modified dextran) CM-dex-WS2 demonstrated a 3.6-fold increase in the adsorption rates for the FAM-DNA probe. Notably, the adsorption ability of FAM-DNA on CM-dex-WS₂ exceeded its hybridization energy for miR-29a, preventing favourable desorption (Fig. 8C). Contrary to this, TMA-dex-WS2 displayed an appropriate adsorption strength, allowing for favourable desorption with miR-29a after hybridization. This interface modulation enabled sensitive and selective recognition of miR-29a (the Alzheimer's

disease biomarker) in human serum, highlighting its potential for bioassays. While the LOD of 745 pM in human serum is impressive, its reliability for detecting miR-29a in clinical samples is uncertain. Clinical samples may exhibit lower miRNA concentrations or background noise levels, potentially challenging sensitive detection.

To harness the fluorescence characteristics of 2D TMDs in water, a method for simultaneous functionalization and exfoliation of WS2 NSs with B-PVA (boronic acid-modified poly(vinyl alcohol)) in water through pulsed sonication was devised [171]. The resultant B-PVA-WS₂ nanosheets were employed to detect glycated hemoglobin (HbA1c). B-PVA-WS2 NSs displayed fluorescence quenching due to HbA1c presence, providing a clear distinction from PVA-WS2. Based on specific fluorescence quenching, B-PVA-WS2 selectively detected HbA1c at low levels of 3.3×10^{-8} M. Although the biosensor showed selectivity for HbA1c in glucose presence, potential interference from other biomolecules in biosamples must be thoroughly evaluated to avoid false-positive or false-negative results. Li and his group enhanced the photoresponse of MoS₂ by using the WS₂@PDA composite [172]. This composite, characterized by the solid electron donor and matched energy band properties of PDA, effectively reduced electron-hole pair recombination (depicted in Fig. 8D). By utilizing the improved photoactivity of WS2@PDA and MoS2, they crafted a sensitive and straightforward PEC biosensor for detecting 5-formylcytosine (5 fC-DNA) formation, with a broad linear range from 5000 to 200 nM and an LOD of 3.7 pM. It demonstrated a stable photocurrent response (RSD 2.26 %) upon repeated irradiation every 20 s, indicating long-term detection stability. Additionally, six parallel biosensors showed good reproducibility with an RSD of 1.87 %. The biosensor could also selectively discriminate between 5-hydroxymethylcytosine, 5-methylcytosine, and cytosine, increasing its specificity. Biological variability, like genetic diversity and sample heterogeneity, along with dynamic DNA modifications under different physiological conditions, may affect the detection of 5 fC-DNA. Standardizing sample preparation protocols and validating them across diverse biological contexts are crucial steps to mitigate this limitation.

Although the unique physicochemical properties of polymerfunctionalized 2D TMDCs have demonstrated their effectiveness in biosensing [164,173], the synthesis of these materials lacks consistency. There is no standardized technique to ensure well-controlled dispersity, thickness, dimensions, and uniform functionalization. Current large-scale synthesis methods result in varied size and thickness distributions, while bottom-up synthesis faces difficulties in achieving well-functionalized nanosheets with controlled thickness. The cytotoxicity of such TMDCs in biosystems appears relatively low, but toxicity in non-targeted areas remains poorly studied and is crucial for potential clinical trials. Regardless of being in the early stages, the functionalization of TMDCs for biomedical use holds promise for highly sensitive biosensing, offering advantages over existing biodegradable materials. However, further research and development are necessary to establish polymer-functionalized 2D TMDCs as a viable nanomaterial for biomedical applications.

4.6. 2D metal-organic frameworks

Metal-organic frameworks (MOFs) are a novel category of porous coordinate materials with properties similar to 2D materials. At the same time, they also possess distinctive features like significant porosity (ranging from 0.5 to 2.5 nm) [174], high adjustability in structure [175], and biodegradability [68], making them suitable for diverse biomedical uses. The fundamental framework of MOFs mainly comprises transitional metal ions connected to negatively charged organic components through various functional links like cyanide, glutamate, formates, triazole, and sulfonates, forming a network through coordination bonds, hydrogen bonding, electrostatic interactions, or π - π stacking [176,177], Fig. 9A. This versatility in metal coordination and organic linkers allows MOFs to create well-organized pores of

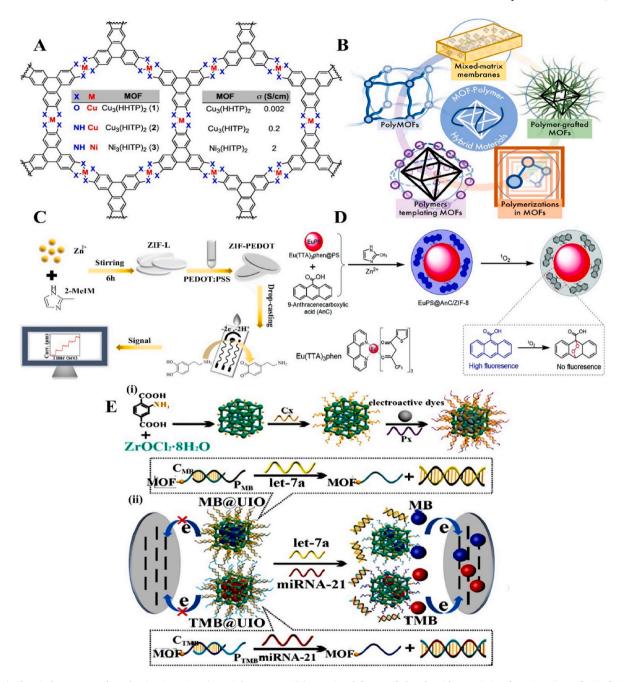


Fig. 9. A) Chemical structure of conductive 2D MOFs (Cu/Ni₃[HHTP/HITP]₂). Reprinted from Ref. [177]. with permission from *American Chemical Society*. B) Systematic depiction of MOF-polymer hybrids focusing on mixed-matrix membranes (MMMs), polymer-grafted MOFs, polymerization in MOFs, polymer templating in MOFs, and polyMOFs. Reprinted from Ref. [189]. with permission from *American Chemical Society*. C) Fabrication of ZIF-L/PEDOT biosensor and the electrochemical detection mechanism for dopamine. Reprinted from Ref. [192]. with permission from *Royal Society Chemical*. D) Synthesis process of EuPS@AnC/ZIF-8 and its operational concept in ratiometric sensing of ¹O₂. Reprinted from Ref. [193]. with permission from *Royal Society Chemical*. E) i) Procedure for fabricating nucleic acid-functionalized MOFs, ii) Explaining the principle behind MOF-based homogeneous electrochemical biosensor for multiple miRNA detection. Reprinted from Ref. [196]. with permission from *American Chemical Society*.

customized sizes and functionality, offering vast potential across numerous technological fields [178,179]. MOFs have gained recognition as a potential sensing material due to their excellent mass transfer capabilities within their carefully designed porous structure [180]. One illustrative case involves zeolitic imidazolate frameworks (ZIFs), which, owing to their substantial porosity and expansive surface area, are thought to enhance the concentration of dehydrogenase enzymes for in vivo electrochemical monitoring of neurochemicals like glucose [181]. Nevertheless, the limited conductivity, reduced selectivity, absence of customizable modification sites, and instability in aqueous

environments constrain the broader utilization of MOFs in biosensor applications [182].

One effective strategy to overcome these challenges involves combining MOFs and conductive polymers to create a single nanostructure with advantageous characteristics of both material categories [183–185]. MOFs offer distinct benefits such as precise porosity, well-defined crystal structures, and controlled metal content. In contrast, conductive polymers contribute qualities like environmental durability, excellent compatibility with biosystems, large surface area, ease of processing, robust mechanical and chemical resilience, and suitability for biosensor applications [186,187]. Research groups have explored a broad range of approaches to create MOF-polymer composites, such as polymer-templated MOF growth, polymer-grafting particles, and MOF preparation with polymer ligands (polyMOFs, as shown in Fig. 9B). The controlled incorporation of MOFs and polymers frequently results in the creation of composite materials that possess either novel characteristics or a combination of properties that surpass those of a mere physical blend of individual components [188]. Due to the synergistic interaction of MOFs and other active elements, these composites are suitable for diverse applications, encompassing storage and separation processes, heterogeneous catalysis, energy devices, biosensing technologies, and biomedical applications [189,190].

To expand the conductivity and utility of MOFs in biosensors, an efficient approach to integrating Co-Ni(Fe)-MOF NSs with conductive PPy for detecting glucose through non-enzymatic electrochemistry was devised [191]. The modified electrode with Co-N(Fe)-MOF/PPy nanocomposite offered exceptional sensitivity (1805 µAmM⁻¹cm⁻²) and a LOD of 1.13 µM for glucose detection, attributed to its synergistic active sites and excellent electron transfer capabilities. The glucose sensor also exhibited high selectivity for glucose over various interferences, with excellent reproducibility and stability. It is a promising platform for sensitive glucose detection in actual samples, achieving recoveries between 103.7 % and 109.5 %. The researchers suggested that this approach opens possibilities for creating various composites of MOFs and conducting polymers to enhance electrochemical biosensing. Yet, the study lacks a comprehensive analysis of nanocomposite's long-term stability and reproducibility in real-world conditions. While short-term stability is demonstrated, further investigation into durability under varied environmental conditions, such as temperature fluctuations and exposure to different analytes, is needed to assess practical applicability thoroughly.

To address issues such as poor stability, poor catalytic activity, and low conductivity in the water dielectric medium of leaf-shaped ZIF, Liu and colleagues introduced the conductive polymer (PEDOT: PSS) into the ZIF-L for sensitive sensing of DA [192]. The resultant ZIF-L/PEDOT composite not only maintained the ZIF-L surface structure but also displayed impressive electrochemical properties due to improved electron transport at the ZIF-L and PEDOT: PSS interface (Fig. 9C). The established biosensor exhibited a LOD of 7 nM and a broad linear range from 25 to 500 μM for DA. Furthermore, the biosensor demonstrated excellent long-term storage stability, retaining approximately 90.52 % of its original current after 28 days of repeated testing. These findings highlighted the latent effect of ZIF-L/PEDOT on disease detection. The study mentioned a slight decrease in sensitivity over time due to the detachment of excess ZIF-L/PEDOT particles from the electrode surface during prolonged immersion. Developing strategies to enhance the stability of the electrode and prevent particle detachment could improve long-term performance. Additionally, fluorescent MOFs with the polymer can detect ions, gases, and vapors. For instance, the combination of 2-methylimidazole, 9-anthracenecarboxylic acid (AnC), zinc nitrate, and europium chelate-doped polystyrene (EuPS) spheres in a methanol solution can result in the creation of EuPS@AnC/ZIF-8, a dual-emissive material with excellent fluorescence [193]. The fabrication process of the sensor involved the self-assembly of EuPS, AnC, and ZIF-8, resulting in a core-shell structure where EuPS served as a reference signal emitter. At the same time, AnC acted as a response signal emitter for singlet oxygen (1O2) detection (Fig. 9D). The integration of AnC and EuPS within the ZIF-8 host not only retained their original fluorescence properties but also prevented the leakage of AnC, enhancing the sensor's stability and performance. This biosensor detected ¹O₂ due to its ability to emit different fluorescent signals of Eu³⁺ and AnC to ¹O₂. It is sensitive, ranging from 0.5 to 200 μM and having a LOD of 43 nM (S/N = 3). It could distinguish ¹O₂ from other reactive oxygen species. Notably, the color change of the sensor in response to stimuli is readily observable with the naked eye under a UV lamp. The biocompatibility test performed using HeLa cells is an excellent initial step. Additional studies on

various cell types and tissues would provide a more comprehensive understanding of the sensor's behaviour in biosystems. Furthermore, long-term stability studies, particularly under physiological conditions, are crucial. This research offers new insights into designing multifunctional composites using MOFs.

Functionalizing MOFs with natural biopolymers, such as proteins, nucleic acid (dsDNA, ssDNA, and aptamer), and cellulose, has been a significant area of recent research. MOF-based nucleic acid biosensors offer distinct advantages over biosensors based on graphene, GO, and AuNP [194]. These advantages include a conjugated π -electron system, open metal sites, flexible porosity, and high loading capacity, making them well-suited for detecting biomolecules, bacteria, and cells [195]. An innovative electrochemical biosensor capable of susceptible detection of two miRNAs was developed, showing potential as valuable cancer biomarkers for diagnosing and treating tumors [196]. They achieved this by attaching functionalized dsDNAs onto MOF surfaces through adsorption and encapsulating electroactive dyes (MB and TMB) within MOF cavities. When target miRNAs hybridized with probe DNAs, the electroactive dyes released from the MOFs, enabled the concurrent analysis of miRNA-21 and Let-7a, both classified as miRNAs (Fig. 9E). The biosensor exhibited high sensitivity, with LODs of 3.6 fM for let-7a and 8.2 fM for miRNA-21 in the ranges of 0.01-10 pM and 0.02-10 pM, respectively. Real sample analysis of serum from breast cancer patients further validated the biosensor's potential for accurate and early cancer diagnosis, detecting concentrations of 19.68 nM for let-7a and 23.24 nM for miRNA-21. However, the synthesis process described for preparing the dsDNA-capped MOFs involves several steps and reagents, which may limit its scalability and practicality for large-scale production. Simplifying the synthesis protocol could enhance the feasibility of widespread adoption.

Polymer integration with MOFs facilitates solvent interactions, improving dispersion, assembly, and processability while enhancing biocompatibility and conductivity for biosensing applications. However, challenges remain. Optimizing polymer loading while preserving porosity is essential, as is understanding the homogeneity of polymer guests within MOF templates. Factors such as MOF crystal size and polymerization rate affect composite performance but lack systematic study. Additionally, the influence of polymers on mass transport within microporous MOFs needs clarification. Understanding MOF-polymer interactions, polymer arrangement in pores, and structural impacts is crucial for performance optimization [197]. Advanced characterization techniques, like solid-state NMR and computational modelling, offer insights into MOF-polymer interactions and arrangement, guiding synthetic procedures for enhanced composite performance. Integrating experimental studies with computational methods holds promise for addressing experimentally challenging questions, such as MOF stability and polymer arrangements within pores. Lastly, future research should focus on scaling up the production of polymer-functionalized 2D MOFs using low-cost synthesis methods while maintaining control over their size, shape, and porosity [198].

5. Emerging functionalized 2D materials for biosensor and bioelectronics applications

Biosensors and bioelectronics that utilize polymer-functionalized 2D materials have found diverse applications across several domains, such as food quality and safety, environmental monitoring, disease biomarkers, and electronic skin. These applications demonstrate the immense potential of such materials in advancing healthcare, environmental protection, and safety. Table 2 presents a detailed comparison of biosensors based on functionalized 2D materials in spiked real samples.

5.1. Food safety and quality monitoring

Food consumption is a fundamental human need, and ensuring its quality and safety is vital for maintaining a healthy life. The ongoing

 Table 2

 Comparison of biosensors utilizing polymer-functionalized 2D materials in real samples: Polymer functionality, analytical performance, advantages, and challenges.

Biosensor designs	Role of polymer	Analytes	Linear range	LOD	Real samples/ Recovery	Advantages	Challenges	Suitability	References
Food safety and qua	ality monitoring								
Au NPs/ NCPDs@FLBP/ SPE	Act as a conductive matrix, improved transmission path, carrier mobility, surface area, and molecular adsorption	E. coli O157: H7	$\begin{array}{c} 1.0 \times 10^{-19} - \\ 1.0 \times 10^{-6} \\ \text{mol/L} \end{array}$	$3.33\times\\10^{-20}~\text{mol/L}$	Chicken meat	Portable electrochemical system, smartphone control, reasonable specificity, sensitivity, and reproducibility, long term stability (up to 30 days)	Optimization of experimental conditions, sensitivity to environmental conditions, and interference	In vitro	[200]
CTAB-MoS ₂ -NSs	Stabilization of exfoliated MoS ₂ , prevented restacking, facilitated smooth film formation	S. typhimurium	$10^1 - 10^7$ $CFUmL^{-1}$	10 CFU mL ⁻¹	-	Superior electron transport, sensitivity, and specific binding	Insufficient incubation time, limited sensitivity for low concentrations	In vitro	[201]
MNPs/CS/g-C ₃ N ₄	Provided a biocompatible surface for enzyme immobilization, enhanced stability, and reusability	Lactose	0.9–100 mm	0.3 mm	Dairy products spiked/un-spiked	Cost-effective, highly active site, homogeneity, metal-free composition, good magnetism, high specificity, short-term stability (up to 7 days), and sensitivity (4.4 μ A mM $^{-1}$)	Uniform dispersion of nanoparticles, optimization of conditions, short long-term stability	In vitro	[205]
PPy-GO-AuNPs	Immobilized GO and Au NPs on electrode surface, enhanced electrocatalytic activity and sensitivity	H ₂ O ₂	2.5 – 25 mM	5 μΜ	Milk/96.0-100.6 %	Fast electrocatalytic activity, rapid response (2 s), high sensitivity (41.35 μ A μ M ⁻¹ cm ⁻²) at S/N = 3, stability (up to 30 days)	pH effect on electrochemical oxidation, limited LOD	In vitro	[207]
NiONSs/PANiNW/ GO	Enhanced homogeneity, active surface area, electrical conductivity, selectivity, and sensitivity	Glucose	2 μM-5.560 mM	0.5 μΜ	Fetal bovine serum/ 99.2–100.4 %	High selectivity, sensitivity (376.22 μA mM^{-1} cm $^{-2}$ at S/N $=$ 3, and electrocatalytic activity	Complex fabrication process	In vitro	[206]
PDA-coated Cu- MOF	Acted as a signal amplification marker, enhance sensitivity	AFB_1	0.05–2 ng/mL	0.01 ng/mL	Lotus seed, maize/ 80.3–104.6 %	Improved sensitivity, cost-effective, simple, avoidance of mAbs loss, excellent hydrophilicity, and biocompatibility, stability at room temperature (up to 90 days) and at 45 °C (up to 7 days)	Stability and bioconjugation capacity of hybrid, optimization of concentration and pH of DA solution		[203]
Ti ₃ C ₂ T _x /PVDF	Acted as an efficient electrochemical redox probe, improved binding sites	OTA	$1.0 \text{ fg mL}^{-1} - 1.0 \text{ ng/mL}$	$2.15~\mathrm{fg}~\mathrm{mL}^{-1}$	Grape juice/92–104 %	Excellent electroactive surface sites, selectivity, reproducibility, and stability (up to 7 days)	Optimization of experimental parameters	In vitro	[204]
Environmental mon graphene/PANI	nitoring Improved electron transfer kinetics, surface-to-volume ratios	Zn ²⁺ , Pb ²⁺ , Cd ²⁺	1 –300 μ g L^{-1}	Zn ²⁺ : 1.0 μg L ⁻¹ , Cd ²⁺ , Pb ²⁺ : 0.1 μg L ⁻¹	Human serum/ 93.8–109.7 %	Low cost, disposable, enhanced sensitivity, and electrical conductivity	Agglomeration of graphene at higher concentrations, low stability	In vitro	[211]
PM/g-C ₃ N ₄	Improved electron transfers, electrocatalytic activity, and sensitivity	Cd^{2+} , Pb^{2+}	0.1–1.0 μΜ	Cd ²⁺ : 0.02 μM, Pb ²⁺ : 0.008 μM	Real water/ 92.5–106.4 %, 90–102.6 %,	Low-cost, efficient adsorption, sensitivity, and stability (up to 7 days)	Selective detection, limited LOD	In vitro	[212]
Fe ₃ O ₄ @Au-PPy/ GO	Enhanced electron transfer rate and stability	Triclosan	$\begin{array}{c} 1.0 \times 10^{-8} - \\ 1.0 \\ \times 1.0^{-6} \ \text{M} \end{array}$	2.5×10^{-9} M	Urine/98.0 %, Toothpaste/101.0 %, Shampoo/96.6 %, Liquid soap/101.6 %	Enhance effective surface area, electrochemical reaction, and storage stability (up to 14 days)	Complex fabrication process, optimization of nanocomposite film	In vitro	[213]
AChE/Con A/ PDA/RGO-GNP	Reduced GO and GNP under alkaline conditions, improved electrical conductivity and protein immobilization	Carbofuran	5–40 mg/kg	0.012 mg/kg	Tomatoes/101 %	Simple approach, low cost, specific binding affinity, improved electrochemical activity, surface area, and sensitivity	Balance between enzyme binding size and pesticide, blockage of AChE active sites	In vitro	[216]
g-C ₃ N ₄ /CMCS/ ZnCdS/ZnS QDs	Acted as ligand, improved biocompatibility, and fluorescence	Mn ²⁺	1.961×10^{-6} - $1.667 \times$ 10^{-5} mol/L	-	Real water/90–107 %	Simple approach, eco-friendly, long- term stability, and reproducibility	Stability of nanocomposites to surrounding environment	In vitro	[247]

Detection of disease biomarkers

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Table 2 (continued)

Biosensor designs	Role of polymer	Analytes	Linear range	LOD	Real samples/ Recovery	Advantages	Challenges	Suitability	References
MXenes-Apt2/ exosomes/Apt1/ PNIPAM-AuNPs	Enhanced aptamer immobilization and catalytic ECL signals	MCF-7	5×10^2 – 5×10^6 particles μL^{-1}	$\begin{array}{c} 125 \ particles \\ \mu L^{-1} \end{array}$	Serum/95–104 %	High sensitivity, specific recognition, amplification strategy, surface area, and conductivity	Complex fabrication process	In vitro	[219]
Prussian blue- polythiophene- GO	Enhanced electronic transmission capacity and electrocatalytic performance	H_2O_2	$\begin{array}{c} 1.0-100.0 \\ \mu M \end{array}$	0.32 μΜ	Milk/95.0-105.0 %	Large specific surface area, high conductivity, selectivity, and sensitivity (0.1066 μ A/ μ M) at S/N = 3	Experimental parameters optimization	In vitro	[220]
TiO ₂ /Ti ₃ C ₂ T _x /NUF	Acted as electrochemical active sites	DA, UA	DA: 0.002–100 μM, UA: 0.001–60 μM	DA: 0.2, UA: 0.18 nM	Urine/98.4–100.9 %.	Large electrode effective area, catalytic activity, anti-fouling property, anti-interference ability	Heterojunction stability, electrode microstructure optimization	In vitro	[248]
PPY/MXene	Increased electroactive surface area, conductivity, and sensitivity	M. tb	100 fM - 25 nM	11.24 fM	Human sputum/ 90.52–100.8 %.	Cost-effective, high electrochemical activity, sensitivity (21.92 $\mu A~nM^{-1}~cm^{-2}$), selectivity, rapid response (10 s)	DNA probe immobilization, hybridization optimization, limited selectivity	In vitro	[249]
Wearable bioelectr	onics and E-Skins							<u> </u>	
PANI-doped $F-Ti_3C_2T_x$	Enhanced ion-to-electron charge transfer, conductivity, and electrochemical stability	H_3O^+	$pH\approx 1{-}11$	-	Human sweat	Highly selective sensitivity, electrochemical stability, fast response, anti-interference ability	$Ti_3C_2T_x$ oxidation, potential drift in $Ti_3C_2T_x$ / PANI electrodes	In vivo	[224]
WS ₂ /PDMS/G	Acted as a flexible membrane substrate, improved repeatability and stretchability	Water moisture	-	-	Human breath	Sensitivity up to 2357 (for 90 % humidity), fast response time, repeatable moisture sensing properties, transparency, flexibility, stretchability, and real-time monitoring	Optimization of electrode structure, improvement of sensitivity	In vivo	[225]
rGO/PET/textile hybrid	Maximize electrical conductivity, durability, and biocompatibility, mitigate polarization effects, and limit ion intercalation	Various biopotentials (EEG, ECG, EMG)	-	-	-	Exceptional biocompatibility, robust, washable textile-based structure, reduced electrode polarization	Laser power optimization, precise integration of rGO into PET, variations in laser power	In vivo	[250]

NCPDs: N-doped carbonized polymer dots; *E. coli: Escherichia coli* O157: H7; S. typhimurium: *Salmonella* typhimurium; MNPs: Magnetic nanoparticles; PVDF: Polyvinylidene fluoride; OTA: Ochratoxin A; GNP: Gold Nanoparticles; PANIPAM: Poly (N-isopropylacrylamide); NUF: Covalent organic polymers; *M. tb: Mycobacterium tuberculosis*; CMCS: Carboxymethyl chitosan; EEG: electroencephalography; ECG: electrocardiography; ECG: electromyography.

globalization of markets is diminishing the prevalence of foods being processed and consumed locally where they are produced. This global integration of the food supply chain necessitates novel approaches to guarantee food safety and quality, as various biotic and abiotic factors compromise food quality during multiple stages. Consequently, routine food safety and quality monitoring at different points in the supply chain becomes imperative. Recently, biosensors have been designed to detect harmful analytes in foods, with sensitivity significantly boosted by the use of 2D materials [199]. Polymer-functionalized 2D materials, with their exceptional physiochemical properties, are ideally suited for incorporation into biosensors and other tools employed in food quality analysis.

For instance, a new portable DNA biosensor was fabricated using NCPDs grown onto a few BP sheets as electrode modifiers, combining methylene blue (MB) as a chemical indicator and Au NPs for immobilizing thiolated ssDNA probes for sensing the specific DNA sequence of E. coli O157: H7 [200], (Fig. 10A). The biosensor exhibited a broader linear range (1.0 \times 10 $^{-19}$ to 1.0 \times 10 $^{-6}$ mol L $^{-1}$), a low LOD (3.33 \times 10^{-20} mol L^{-1} , three σ), and high sensitivity, validated by detecting E. coli genomic DNA in spiked samples of chicken meat with good reproducibility, RSD values of 2.60 % and 4.03 % for two different concentrations of target DNA sequence. It also exhibited long-term stability when stored in a refrigerator at 4 $^{\circ}$ C for up to 30 days. Real-world application potential is highlighted, but challenges like sample complexity and device integration into portable systems need further exploration for practical field use. Addressing power consumption, miniaturization, and user interface design would enhance the feasibility of on-site detection. In another study, CTAB functionalized-MoS₂ nanosheets were produced to detect S. typhimurium on a microfluidics electrode [201]. The nanosheets were fabricated through exfoliation assisted by CTAB, and CTAB-MoS₂-NSs acquired a positive charge, with subsequent deposition on a patterned hydrolyzed ITO microelectrode using electrostatic forces (Fig. 10B). This integrated setup was then combined with a PDMS microfluidic device, resulting in the formation of a lab-on-a-Chip system designed explicitly for S. typhimurium recognition. The device exhibited a high sensitivity of 1.79 k Ω /CFU $^{-1}$ mL cm $^{-2}$ and an LOD of 1.56 CFU mL $^{-1}$ in a broad linear range from 10^1 to 10^7 CFU mL $^{-1}$. However, the stability of the immunochip decreased significantly after the 3rd week of storage, with a 14.2 % decrease in impedance signal observed by the end of the 7th week. This limitation could affect the long-term reliability and practical application of the device.

Besides pathogens, these 2D materials are actively utilized for precise mycotoxin detection in food. Mycotoxins are toxic compounds produced by certain fungi on agricultural commodities, such as nuts. grains, and fruits. These toxins contaminate food and feed, posing health risks when ingested [202]. A recent development includes a strip lateral flow immunoassay (SLFIA) designed for on-site analysis of AFB1 in food samples. This innovative approach used PDA-coated Cu-MOF as a signal amplification marker [203]. Under optimal conditions, the modified SLFIA demonstrated a low LOD of 0.01 ng/mL and a broad concentration range of 0.05-2 ng/mL for AFB1. Its application in lotus seed and maize samples yielded reasonable recoveries of 80.3 %-104.6 %, confirming its accuracy in real-world scenarios. Furthermore, stability tests revealed that the SLFIA strips remain effective for up to 90 days at room temperature and tolerate storage at 45 °C for seven days, highlighting their suitability for field use without cold chain transport. However, the cost of materials and equipment required for the SLFIA method, like

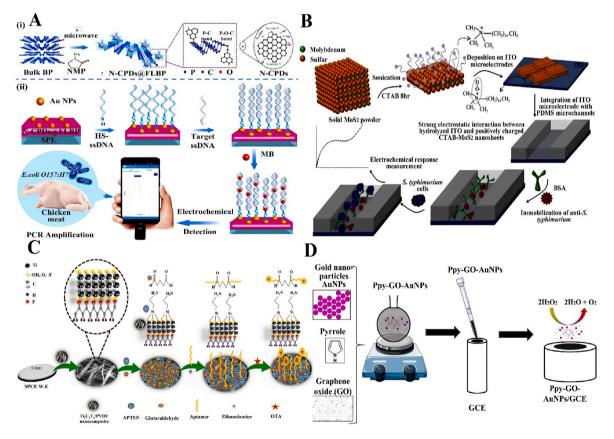


Fig. 10. A) i) One-step synthesis of N-CPDs@FLBP heterostructure, **ii**) Schematic illustration of a portable intelligent electrochemical DNA biosensor for detecting E. coli O157: H7. Reproduced from Ref. [200]. with permission from *Elsevier*. **B)** Diagram illustrating the sequential process of crafting a microfluidic biosensor for detecting S. typhimurium. Reprinted from Ref. [201]. with permission from *Elsevier*. **C)** Systematic depiction of fabrication and working mechanism of aptasensor. Reprinted from Ref. [204]. with permission from *Elsevier*. **D)** Synthesis process of PPy-GO-AuNPs nanocomposites. Reprinted from Ref. [207]. with permission from *Elsevier*.

MOFs, PDA, and monoclonal antibodies, may limit its accessibility, particularly in resource-limited regions or for small-scale food producers. Here, a very sensitive electrochemical aptasensor was fabricated using electrospun Ti₃C₂T_x/PVDF nanocomposite on a screen-printed carbon electrode to detect OTA accurately [204]. Structural and electrochemical analyses revealed enhanced electroactive sites due to the synergistic effect between β-phase PVDF and Ti₃C₂T_x, along with redox probe effective covalent bio-functionalization with aptamer (Fig. 10C). This synergy amplified signal strength and transduction, notably increasing sensitivity with a low LOD of 2.15 fg mL⁻¹ and a broad dynamic range from 1.0 fg mL⁻¹ to 1.0 ng/mL. It had the potential to detect fg mL⁻¹ concentrations of OTA in grape juice, outperforming HPLC with recoveries ranging from 92 % to 104 % and RSDs from 2.99 % to 6.89 %, highlighting its practical applicability and precision in the food industry. Further validation using a broader range of food matrices and comparison with established analytical methods (e.g., HPLC) would strengthen the credibility and reliability of the proposed sensor for real-world applications.

Recent biosensor advancements have also made assessing lactose and glucose levels in food samples easy. For example, an efficient lactose sensor was developed using an MNPs/CS/g-C₃N₄ composite to support cellobiose dehydrogenase immobilization and as a modifier for GCE [205]. The sensor exhibited excellent performance for detecting lactose with a low LOD (0.3 mm), wide range (0.9-100 mm), and high sensitivity (40.8 μA mm⁻¹). Moreover, it demonstrated a rapid response time of 5s. It also proved effective in quantifying lactose in real dairy products, demonstrating its capability for real samples. Although it revealed promising short-term stability over a week, the observed decrease in response after four weeks suggested potential enzyme loss during storage. Enhancing the long-term stability of biosensors would be crucial for their practical utility in real-world applications where extended storage may be required. Zhuang and his team introduced a new glucose biosensor using a hybrid composite of NiONPs, PANi nanowires, and GO on a GCE [206]. Electrochemical studies revealed the superior electrocatalytic activity of NiONSs/PANiNW/GO modified electrodes for glucose oxidation in an alkaline solution, attributed to the presence of PANi. The biosensor demonstrated high sensitivity (376.22 μA mM⁻¹ cm $^{-2}$), a wide linear range (2 μ M-5.560 mM), low LOD (0.5 μ M), and excellent selectivity of glucose against interfering reagents (AA, DA, UA). Real sample analysis of fetal bovine serum showed accurate glucose determination with high recovery rates (99.2 %-100.4 %) and low RSD (1.3 %–1.9 %), highlighting its potential in various practical applications, including clinical diagnosis and food analysis. An extensive investigation of the long-term stability and durability of the sensor can determine its practical utility.

Biosensors can also detect H₂O₂, which is paramount in signalling processes associated with diverse diseases. The ability to accurately analyze H₂O₂, qualitatively and quantitatively, is crucial in sectors ranging from healthcare to industrial processes and food applications. A research group successfully engineered an extremely sensitive electrochemical biosensor to recognize H₂O₂ in food samples. They fabricated the sensor by employing PPy-GO-AuNPs nanocomposites prepared via in-situ polymerization with FeCl₃·6H₂O as an oxidizing agent [207] (Fig. 10D). The resulting PPy-GO-AuNPs/GCE exhibited significant electrocatalytic activity for detecting H_2O_2 , with an LOD of 5 μM and a superior sensitivity of 41.35 $\mu A/mM$ within a linear range (2.5 μM –25 10⁻³ Mm). The practical utility of this nanocomposite was verified in actual milk samples with up to 100.6 % recovery, showcasing its potential for wide-range and sensitive H2O2 detection in healthcare and food analysis. The study also investigated the effect of pH on H₂O₂ determination, with pH 7 identified as the optimal condition for electrochemical analysis with maximum sensor performance. However, exploring the sensor's response across a broader pH range could provide additional insights into its pH sensitivity and applicability in diverse environments.

5.2. Environmental monitoring

Ensuring the safety and well-being of humans and the environment on our planet is fundamentally dependent on continuously monitoring our surroundings, safeguarding human health, overseeing chemical processes, and upholding air and water quality. It also entails the detection of gases, infrared radiation, contaminants like organic and inorganic compounds, pesticides, heavy metals, antibiotics, poisonous gases, bacteria, and more, whether they are present in water or the atmosphere. The existence of these harmful substances poses a substantial risk to both the environment and human well-being. Monitoring and managing these dangerous and toxic elements, radiation, and microorganisms has significantly challenged the scientific community. Regarding this, adopting 2D materials as the basis for sensors has proven highly efficient and is well-suited to modern manufacturing techniques [208,209]. Materials like MOF, BP, and others possess unique properties that make them attractive for producing exceptionally sensitive sensors. These properties are a result of the inherent structures of these materials. When 2D materials are modified with various polymers, their sensitivity, selectivity, biocompatibility, and other characteristics are significantly amplified, as Zhang's team devised a nanohybrid called rGO/PEI/Pd This nanohybrid demonstrated peroxidase-simulated activity due to 3,3',5,5'-tetramethylbenzidine (TMB) oxidation. The Hg²⁺ addition significantly enhanced the peroxidase-like activity, leading to intensified TMB oxidation and a deep blue colour in the solution. Utilizing nanohybrid, their Hg²⁺ sensing platforms achieved an LOD of 0.39 nM in a concentration range of 0.1-25 nM for double-distilled water and approximately 1 nM for actual water samples through spectrographic analysis. Importantly, this approach enabled the visual detection of 10 nM Hg²⁺ in serum and wastewater samples, showcasing potential applications in portable and straightforward environmental sample analysis for low concentrations of Hg²⁺. Although the LOD for Hg²⁺ is reported to be low (0.39 nM), it is essential to consider the potential impact of the sample matrix and interference from other substances, which may affect the sensitivity and accuracy of detection in real-world samples.

The global problem of heavy metal ions (HMIs) pollution is a persistent concern. Consequently, the surveillance of HMIs in natural water sources is crucial and demands efforts to safeguard both environmental well-being and human safety. To address this issue, it is essential to identify these HMIs first. A graphene/PANI-modified electrode was developed using drop casting and electrospray techniques [211]. This electrode was an electrochemical sensor capable of detecting Zn²⁺, Pb²⁺, and Cd²⁺ simultaneously. Compared to the unmodified electrode, the modified electrode exhibited enhanced peak anodic current and electrochemical conductivity (Fig. 11A). It exhibited a linear detection range of 1–300 μ g L $^{-1}$, with an LOD of 1.0 μ g L $^{-1}$ and 0.1 μ g L $^{-1}$ for Zn $^{2+}$, Cd $^{2+}$, and Pb $^{2+}$, respectively. Notably, this modified electrode displayed careful identification of the target analytes in the existence of other metal ions, i.e., Fe²⁺, Mn²⁺, Fe³⁺, Co³⁺, Ni²⁺, and Cu²⁺. It also effectively detected Zn²⁺, Cd²⁺, and Pb²⁺ in human serum at levels consistent with typical healthy adult concentrations. With recoveries between 93.8 % and 109.7 % and RSD below 9.0 %, the method demonstrated high accuracy and reproducibility, showcasing its analytical reliability for clinical applications. While the study showed the effectiveness of electrodes in human serum, their biocompatibility is not explicitly addressed. Additional research uncovered the distinctive sensitivity of poly(melamine)/graphitic-carbon nitride in detecting Pb²⁺ and Cd²⁺ (Fig. 11B). It comprised an innovative, cost-effective, and highly efficient nano-engineered PM/g-C₃N₄ nanonetwork coated on a pre-activated screen-printed carbon electrode (ASPE) [212]. This modified electrode was designed for the selective, sensitive, and concurrent electrochemical detection of toxic HMIs in environmental water sources. The created PM/g-C₃N₄/ASPE is an environmental sensor for the selective and concurrent electrochemical detection of Cd²⁺ and Pb²⁺ ions via the differential pulse voltammetry (DPV) technique. This sensor

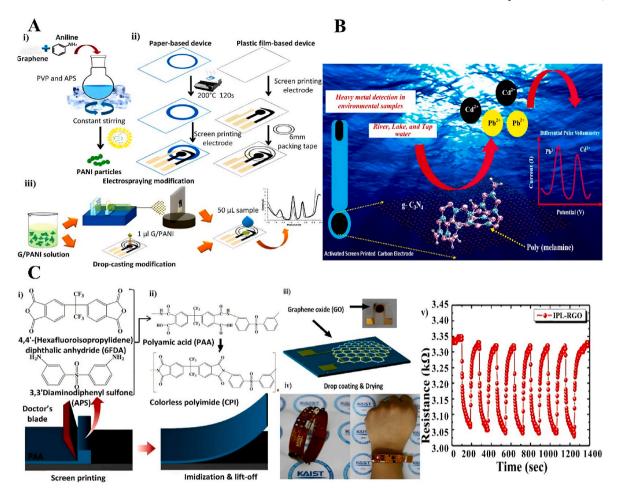


Fig. 11. A) Diagram illustrating **i)** the synthesis of G/PANI nanocomposite, **ii)** the assembly of the electrochemical sensor on plastic film/paper, and **iii)** the drop-casting and electrospray methods. Reproduced from Ref. [211]. with permission from *Elsevier*. **B)** Overview of the process and interaction mechanisms of ASPE-PM/g-C₃N₄ with HMIs in environmental water samples. Reprinted from Ref. [212]. with permission from *Elsevier*. **C)** Illustrations depicting **i)** the synthesis of polyamic acid (PAA), followed by screen printing on a glass substrate with a doctor's blade. **ii)** Subsequent imidization of PAA forms a CPI film, which can be lifted off the glass substrate. **iii)** GO sheets are coated on CPI film with an interdigitated electrodes (IDEs) pattern. **iv)** Optical images of a wearable sensor module with IPL-RGO sensor (blue dotted box). **v)** Resistance changes of the IPL-RGO sensor on the module in response to 20 p.p.m. H₂S at room temperature. Reproduced from Ref. [214]. with permission from *Nature*. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

exhibited outstanding selectivity and sensitivity, with an LOD of 0.02 μM for Cd^{2+} and 0.008 μM for Pb^{2+} . Interference studies revealed selective detection, while analysis of environmental water samples confirmed precise detection within the linear range (0.1–1.0 $\mu M)$ with recoveries of 92.5–106.4 % for Pb^{2+} and 90–102.6 % for Cd^{2+} , indicating its potential as a reliable electrochemical sensor for HMIs. However, the sample preparation process, such as filtration and pH adjustment, may introduce errors or variability, impacting the accuracy and reliability of the analytical results.

There are not only HMIs in the environment but also a reservoir of multiple substances. One of them is bacteria, which can lead to severe illnesses. Saljooqi and co-workers introduced a sensitive biosensor utilizing a GCE enhanced with a magnetic Fe $_3O_4$ @Au-PPy/GO nanocomposite to detect an antibacterial agent, triclosan [213]. This biosensor exhibited improved electron transfer, sensitivity (747.8 μA^{-1} μM^{-2}), linear range (0.01–1.0 $\mu mol\ mL^{-1}$), and a low LOD (2.5 nM) for triclosan. It also showed excellent precision (RSD 4.9 %) and stability (3.25 % decrease after 14 days at 4 °C), making them promising for triclosan detection. While the study reports acceptable sensitivity for triclosan detection, the LOD should be further optimized to meet the requirements of specific applications, especially in trace-level detection scenarios. Moreover, various hazardous gases present in the environment require effective control to prevent potential harm. Optically reduced GO NSs were introduced and fabricated on a highly thermally

stable, transparent, colourless polyimide (CPI) substrate. This result was achieved by subjecting a GO-coated CPI film to intense pulsed light (IPL) irradiation (Fig. 11C). These flexible RGO sheets were gas-sensing layers suitable for wearable systems [214]. The ultrafast IPL irradiation process created RGO NSs on the CPI film in just 4 ms without damaging a plastic substrate. These IPL-RGO sheets displayed significantly enhanced chemical detection capabilities for gases like H₂, H₂S, and C₂H₅OH, with detection ranges and limits of 1-20 ppm and 20 ppm, respectively. In contrast, the original GO sheets showed no response to these gases. These sensing properties remained consistent after subjecting the IPL-RGO NSs on the polymer film to mechanical deformation through 104 bending cycles. While the study demonstrated the IPL-RGO sensor's stability under controlled bending conditions, its performance in real-life situations where the sensor may undergo repeated and unpredictable mechanical stress remains unclear. Assessing its durability in various environmental conditions, such as temperature fluctuations, humidity, and mechanical vibrations, would provide a more comprehensive understanding of its practical utility.

Contamination from pesticides, especially organo-phosphorus ones, is very harmful to people and animals. These pesticides can be found in water bodies and accumulate on fruits and vegetables, posing a potential risk to consumers by affecting the functions of the human body and even causing harm through the inhibition of AChE activity. That is why finding effective ways to detect, identify, and measure pesticides for

human well-being is essential. Electrochemical sensors, using 2D materials and their devices, have been effectively employed to detect pesticides [215]. For instance, Li and colleagues detected carbofuran in tomatoes using an electrochemical biosensor that immobilized AChE on an electrode modified with a con A/PDA-rGO-GNP composite. The biosensor demonstrated a broad detection range of 5–200 mg/kg, with a low LOD of 0.012 mg/kg [216]. Recovery measurements on tomato samples achieved a 101 % recovery for 10 mg/kg and 90 % recovery for 100 mg/kg, with an analysis time of 15–20 min, indicating accurate and rapid determination. Additionally, it demonstrated good repeatability (3.1 % RSD) and stability, retaining 80 % of the initial current after four weeks of storage at 4 °C. However, its specificity to carbofuran in the presence of other compounds commonly found in tomatoes must be thoroughly evaluated.

5.3. Detection of disease biomarkers

A biomarker is a biological signal to predict clinical outcomes, offering a more practical and cost-effective alternative to directly observing challenging endpoints or intermediate outcomes. Clinical biomarkers, applied in disease screening, diagnosis, and monitoring, serve diverse roles, such as prognostic indicators, supporting personalized therapies, anticipating adverse drug reactions, distinguishing cell types, and contributing to pharmacodynamic studies [217]. Achieving highly selective and sensitive detection of biomarkers using bioreceptor-free sensors has proven to be a significant challenge. To address this issue, 2D materials with exceptional properties are utilized,

and their integration with polymers enhances their sensitivity, selectivity, and biocompatibility in disease biomarker detection. For example, early detection of cancer cells can be done by electronic biosensors. Joshi and co-workers developed a portable biosensor for instantaneous sensing of a cancer biomarker called carcinoembryonic antigen (CEA) at room temperature [218]. This biosensor utilized an innovative poly[(1,4-phenylene)-alt-(3,6-(1,2,4,5-tetrazine)/3,6-(1,2,4, 5-dihydrotetrazine))] (PhPTz) as a matrix and rGO as a conducting layer for immobilizing CEA antibodies. The nitrogen-rich semiconducting PhPTz offered advantages such as efficient immobilization of anti-CEA and transportation of charge density variations to the rGO layer (Fig. 12A). As a result, the device displayed a response of $2.75-33.7 \mu A$ across a wide range from 0.25 pg/mL to 0.08 ng/mL for CEA concentrations. Using a modest algorithm, a portable read-out circuit with a voltage divider circuit and an Arduino UNO achieved 100 % prediction accuracy for classifying CEA concentrations. This technology holds potential for POC cancer treatment and diagnosis. The focus on detecting only CEA limits the biosensor's broader applicability. Further refinement and validation of the algorithm for predicting other analytes are needed to ensure accurate real-time monitoring in diverse clinical settings.

By adsorbing MXene on exosome-cultured AuNP/poly (N-isopropyl acrylamide)/GCE biosensors, it is fabricated to detect breast cancerous cells (MCF-7). Initially, these exosomes were grown on AuNP/PNIPAM/GCE by integrating an aptamer with an amino-functionalized layer [219]. Subsequently, MXene-based nanoprobes were synthesized after modifying the surface of MXene electrostatically with many aptamers

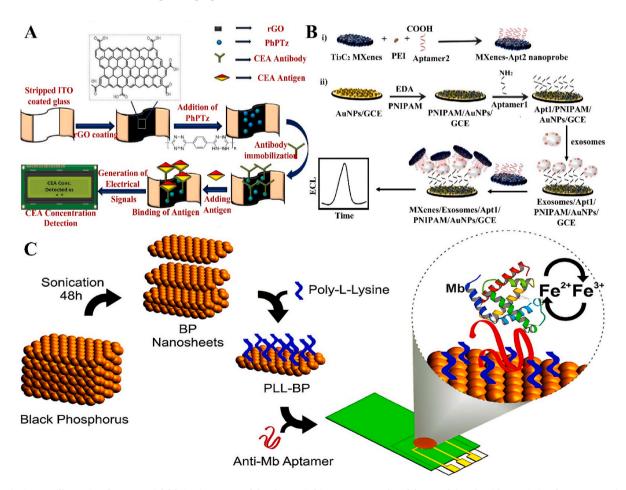


Fig. 12. A) Diagram illustrating the sequential fabrication stages of the electronic biosensor. Reproduced from Ref. [218]. with permission from *Nature*. B) Principle behind the signal amplification strategy in the ECL biosensor for detecting exosome activity. Reprinted from Ref. [219]. with permission from *Elsevier*. C) Illustrated workflow for liquid-phase exfoliation of BP nanosheets and their surface modification to develop a biointerface on an electrode for Mb detection. Reproduced from Ref. [221]. with permission from *American Chemical Society*.

and PEI (as shown in Fig. 12B). Research results specified that the PL intensity of the luminol solution is five times higher for the manufactured nanoprobe compared to pure GCE, supporting the usage of MXene-based aptamer nanoelectrodes for exosome sensors. This system exhibited a detection range of 500 to 5,000,000 particles per microliter and an LOD of approximately 125 particles per microliter. Its reproducibility (RSD: 3.2 %-4.5 %) and stability showcased its potential for clinical serum sample assays. In this study, MXene nanosheets served a dual purpose by catalyzing the ECL process of luminol due to their outstanding catalytic properties and conductivity and offering an increased surface area to efficiently load the aptamer Apt2, thereby facilitating the capture of exosomes. However, the scalability and cost-effectiveness of the biosensor fabrication process need to be considered for widespread adoption in clinical settings. Costly materials or complex fabrication procedures may hinder large-scale production and accessibility.

Several biomarkers cause oxidative stress, which is the root cause of several diseases like neurodegenerative diseases, cardiovascular diseases, kidney diseases, lung diseases, bone, blood, and liver disorders, and many other infectious diseases. Oxidative stress (OS) is defined as a situation in which there is an imbalance between the formation of ROS and the presence of antioxidants that can neutralize them. The degree of OS can be assessed by examining the levels of different biomarkers [19]. According to one study, using an electrochemical technique, a Prussian blue-polythiophene-GO membrane biosensor was prepared to detect the level of one primary OS-causing biomarker, such as H2O2, in a range of 1.0-100 µM [220]. In the case of cardiovascular disease, numerous cardiac biomarkers have shown great interest due to their ability to serve as indicators of cardiovascular incidents. Among these markers, serum cardiac biomarkers, with myoglobin (Mb) particularly noteworthy, hold significant importance in clinical diagnosis. An elevated level of myoglobin signifies myocardial injury and has now established itself as the gold standard for diagnosing acute myocardial infarction (AMI), often known as a heart attack. To address this problem, Vinod and his colleagues reported the electrochemical analysis of the redox-active cardiac biomarker Mb with nanostructured BP electrodes functionalized with aptamers. This process involved the direct measurement of electron transfer [221]. The few-layer BP nanosheets, synthesized for this purpose, were treated with pLL to enhance their interaction with anti-Mb DNA aptamers generated on the nanostructured electrode (Fig. 12C). The resulting biosensor platform showcased an exceptional LOD of 0.524 pg \mbox{mL}^{-1} and high sensitivity of $36~\mu A~pg^{-1}~mL~cm^{-2}$ for Mb. It covered a dynamic response range of 1.0pg mL⁻¹ to 16 μg mL⁻¹ in serum samples, setting the stage for forming POC systems for multiplexed cardiovascular disease diagnosis in complex human samples. Still, the biocompatibility and potential cytotoxic effects of BP nanosheets, PLL, and the functionalized aptasensor need thorough investigation before clinical translation.

5.4. Wearable bioelectronics and E-skins

Wearable bioelectronic devices incorporating 2D materials have been extensively researched. This section explores notable examples of such devices, particularly in skin monitoring, which is an essential application for 2D material-based biosensors. These sensors display a capacity for secure and personalized attachment to the skin, ensuring sustained sensitivity over extended durations. The distinctive interface with human skin presents a valuable research domain for monitoring temperature, pressure, touch, and diseases [15]. Consequently, due to their unique physicochemical advantages, a skin-sensing device (E-skin) incorporating 2D nanomaterials has been developed for detecting various human body indicators, presenting a significant advancement over traditional materials-based E-skins in the past decade [222,223].

The growing focus of today's society on personal health has led to the thriving development of wearable physiological and physical signal monitoring medical devices for treating, diagnosing diseases, and

assessing daily health. By utilizing functionalized 2D materials, these wearable electronic devices can offer high sensitivity, softness, and versatility, enabling precise detection of indicators like sweat acidity, alkalinity, blood glucose concentration, etc. Here, a novel and practical electrochemical sweat pH biosensor was designed to continuously track hydrogen-ion levels in human sweat [224]. A flexible electrode on a PET substrate was crafted using screen-printing technology by incorporating fluoroalkyl silane functionalized $Ti_3C_2T_x$ and PANI membrane. PANI-doped F-Ti₃C₂T_x enhanced responsiveness, pH sensitivity (-41.91 mV per pH), and reversibility due to its superhydrophobic characteristics. When the resulting pH sensor was used to monitor human sweat pH in real-time during exercise, it demonstrated accurate detection on curved surfaces and pH values above 8 for both genders, indicating its versatility and applicability in health monitoring and environmental sensing (Fig. 13A). This study presented a practical solution for online monitoring the pH of human sweat using a reliable MXene-based mini pH sensor. In addition to sweat sensing, polymer-functionalized 2D materials are extensively valuable for humidity sensing. The WS₂ film was a robust high humidity sensor, demonstrating remarkable sensitivity (up to 2357) for 90 % humidity and repeatability in flat and mechanically flexible states when exposed to periodic saturation with moisture gas [225]. Moreover, when integrated into a PDMS membrane, the sensor flexibility allowed for repeatable water moisture sensing even under bent curvature down to 5 mm and withstands up to 40 % elastic stretch without significant performance degradation. Furthermore, a transparent, flexible, and stretchable sensor is developed using graphene electrodes and a thinner PDMS membrane as a substrate. When laminated onto the skin, this sensor showed repeatable humidity responses under compressing, stretching, and relaxing states. Also, real-time monitoring of human breathing revealed its effectiveness in tracking fast (1 s) and slow (5 s) breathing patterns, showcasing significant potential for healthcare applications like real-time dynamic monitoring of human breath. However, further optimization of electrode structures may be required for improved sensitivity, as current levels achieved by rigid and flexible devices are low (nA and pA levels, respectively).

Furthermore, these materials enable the visualization of various physical signals, like pressure and strain. Pressure sensors have been successfully developed using 2D materials with capacitive, triboelectric, and piezoresistive effects. For example, a very flexible, repeatable, and highly sensitive self-powered tactile biosensor based on triboelectric effects was developed with wrinkled PDMS/MXene composite films [226]. The resulting composite film, generated through UV ozone treatment, introduced -OH groups and surface wrinkles on PDMS, while MXene addition introduced -F and O-containing groups, collectively contributing to superior sensitivity of 0.06 V/Pa during 80-800 Pa and 0.18 V/Pa during 10-80 Pa. The results suggested that this sensitivity is higher compared to other self-powered tactile sensors and has the potential for monitoring complex human physiological indications and replicating human touch sensations in E-skin technology (Fig. 13B). Yet, the potential impact of factors such as temperature, humidity, and exposure to contaminants that may influence the performance and sensitivity of sensors needs to be addressed.

From prior research, it is evident that various compressible and elastic carbon aerogels (CECAs) composed of CNTs, graphene, and 2D nanomaterials have demonstrated effective piezoresistive behaviour. These materials are highly suitable for sensitive sensors in wearable Eskins and bioelectronics. A light CECA was developed to address challenges related to insufficient interaction between nanomaterials by employing bacterial cellulose (BC) as a nano-binder to link Ti_3C_2 nanosheets [227]. The resulting CECAs exhibited excellent elasticity and compressibility (up to 99 % strain for over 10,000 cycles), as shown in Fig. 13C. In contrast, the device demonstrated high sensitivity (12.5 kPa $^{-1}$) over a broad working pressure range (0–10 kPa) and good linearity, reaching 95 % workable strain. Notably, the sensor displayed sensitivity to subtle pressure and strain changes, making it capable of

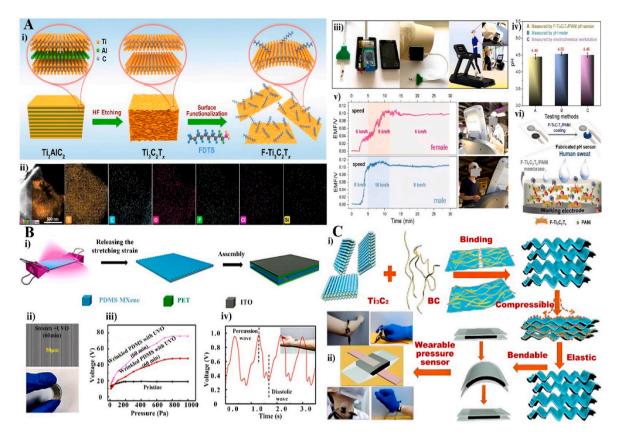


Fig. 13. A) i) Preparation and surface modification of $F-Ti_3C_2T_x$ NSs. ii) TEM image and element mappings of uniform Ti, C, O, F, Cl, and Si distribution in NSs. iii) Wearable pH-sensing device image with a Li-ion battery, mini potentiometer, $F-Ti_3C_2T_x/PANI$ pH sensor, and sports cap for immobilization. iv) pH sensor for continuous real-time monitoring of sweat pH. v) Comparison of human sweat pH measured by different methods. vi) $F-Ti_3C_2T_x/PANI$ compound systematic functioning as an H_3O^+ -selective membrane and intermediate solid contact layer in SC-ISEs for sensing human sweat pH. Reproduced from Ref. [224]. with permission from *American Chemical Society*. B) i) Fabrication of wrinkled PDMS/MXene composite film. ii) SEM images of PDMS surface and cross-section and flexibility demonstration (PDMS/MXene). iii) Performance between PDMS/MXene and PDMS. iv) Real-time V-t curve of tactile sensor demonstrating pulse in action. Reproduced from Ref. [226]. with permission from *Elsevier*. C) i) C-MX/BC-x carbon aerogel synthesis with bendable ability. ii) C-MX/BC-2 applications in biosignals detection, plus the assembly of a sensor (center). Reproduced from Ref. [227]. with permission from *Elsevier*.

recording variations in human speech when attached to the throat. When placed on the human throat, the sensor captured electrical changes while uttering words like "carbon" and "super." Scientists anticipated that these sensors on CECA, with their ultra-high linear sensitivity, superior mechanical properties, and low LOD (1.0 Pa), hold significant promise for future applications in wearable sensing devices. Although it demonstrated remarkable fatigue resistance and structural stability under cyclic compression, its long-term durability under real-world conditions, such as exposure to moisture, temperature variations, and mechanical wear, needs further investigation.

6. Challenges and future directions

Recent advancements in 2D materials have found widespread use, especially in biosensors and bioelectronics. Numerous studies have shown that new 2D materials with polymer functionalities can be used in bioelectronic devices. However, there is still a long way to go before they can be used in real life.

The primary challenge lies in optimizing the synthetic process of polymer-functionalized 2D materials for practical applications, where factors like uniformity, morphology, pore diameter, and surface area significantly impact their properties. Complex synthetic methods for bioelectronic devices pose a risk of uncontrollability during production expansion, hindering ideal structural features and specialized functions. The stability of materials prepared through such methods raises concerns, potentially causing structural decomposition and collapse in complex environments. Moreover, for the successful commercialization

of bioelectronic devices, it is essential to have cost-effective manufacturing methods that allow for high throughput and large-scale customization. Traditional manufacturing approaches like etching processes, photolithography, and high-vacuum deposition require expensive equipment and are time-intensive, leading to higher product costs. Advanced techniques like printing offer the potential for mass production, uniform nanosheet sizes, and reduced manufacturing expenses. Additionally, achieving favourable biocompatibility requires that the fabrication process avoid using toxic materials or solvents. Innovative manufacturing technologies are crucial to ensuring that polymerfunctionalized 2D materials are compatible with processing parameters.

Second, the interactions between biological entities and polymerfunctionalized 2D nanomaterials are less covered. Understanding how these materials influence signal pathways and control biological processes, including interactions with structural proteins and genetic materials, can open various opportunities to discover biosensing applications and enhance the current LOD. This knowledge deepens our understanding of how functionalized 2D material characteristics, such as size, shape, and chemical composition, impact specific biological responses, biocompatibility, and cytotoxicity. Although the endocytosis pathways and biological activities of specific nanosheets in cancerous cells have been explored, such studies are in their infancy, and more endeavours are required to delve into the interactions of various polymer/2D materials with biomolecules. For instance, Tao et al. [228] introduced a doxorubicin hydrochloride-loaded delivery platform utilizing BP nanosheets. Their work innovatively uncovered the endocytosis pathways as well as the biological functions of PEGylated BP nanosheets, illustrating the transportation route via 'micropinocytosis,' 'late endosomes,' and 'lysosomes.' with autophagy playing a role in the breakdown of PEGylated BP nanosheets.

Third, the long-term stability of polymer-functionalized 2D materials and their application in sensing platforms have been underexplored in existing research. Stability is crucial in developing sensing devices, particularly those intended for continuous and extended analyte monitoring. Achieving long-term stability necessitates a combination of characteristics such as high skin compatibility, environmental resilience, strong binding properties, mechanical robustness, and flexibility. Although efforts have been made to enhance the functionality of 2D materials with polymers to meet these requirements, the stability of these composites tends to diminish over time, an aspect that remains inadequately investigated. Addressing this challenge involves a strategic approach to polymer selection and utilization, encompassing factors such as polymer type, functionalization methods, and thickness. The ability to tailor the stability of polymer and 2D materials is crucial for their diverse applications. For instance, reusable devices must endure repeated attachment and detachment cycles to ensure consistent performance. Therefore, future research endeavours should prioritize investigating the long-term stability of these materials to develop durable sensing platforms with extended lifespans.

Fourth, each emerging 2D material possesses unique advantages, yet its limitations cannot be overlooked. While polymers have significantly enhanced the properties of 2D materials for biosensing applications, their widespread application in clinical diagnosis, particularly for POCT, remains challenging. Challenges include miniaturization of instruments, efficient and stable loading between nano-carriers and targets, and probe cytotoxicity or interference. Given the suitability of simple and convenient biosensors for POCT, the design of automated sample devices and portable instruments is crucial. Hence, novel approaches leveraging nanotechnology and signal amplification strategies are imperative for practical and accurate clinical applications. For example, creating hybrid nanomaterials with diverse materials, including but not limited to 2D materials, proves effective in avoiding inherent shortcomings. Combining the benefits of $\text{Ti}_3\text{C}_2\text{T}_x$ NSs with efficient charge transfer and WSe₂ NSs with abundant active sites, a Ti₃C₂T_x/WSe₂ hybrid biosensor for volatile gas sensing is meticulously engineered [229]. The resulting hybrid material forms numerous heterojunction interfaces, significantly increasing the adsorption of oxygen species and consequently releasing more electrons back to the channel, thereby improving the detection sensitivity of oxygen-containing volatile organic compounds by over 12 times. Moreover, material hybridization can yield intriguing material interfaces, enhancing recognition ability and signal amplification while mitigating material defects. Furthermore, recent years have also seen remarkable advancements in nanotechnology, with numerous 2D materials being employed in modified signal amplification strategies and sensitivity, such as nucleic acid signal amplification, multivalent aptamer, and dual-signal probes. These methods significantly enhance signal throughput, speed, sensitivity, and simplicity and reduce sample consumption, enabling the detection of biomolecules at the single-cell or single-molecule level, with promising potential for practical applications.

Lastly, getting functionalized 2D material-based biosensors to find biological or chemical molecules on the market is much harder than getting activity-tracking devices or regular lab-based biosensors on the market. This complexity arises from the need for continuous biosensing and accurate measurement of these molecules. Here's a detailed breakdown of the key considerations and potential solutions:

 A reliable power supply is crucial for the robust performance of bioelectronics. However, conventional power sources tend to be bulky and rigid, and they may contain hazardous materials that are not ideal for soft, curvilinear, and limited-space biological organisms. Solutions such as lithium-ion and lithium-polymer batteries,

- harvesting methods, biofuel cells, kinetic solar cells, and wireless power transfer methods are viable for meeting energy needs.
- 2. Biosensor devices generate massive amounts of data during experimental testing. Future compatibility with cloud technology and AI could enable medical professionals to monitor patients remotely, allowing them to recuperate at home. Advanced wireless communication technologies like Wi-Fi, NFC, Zigbee, Bluetooth, body area networks, and new protocols such as LBE and 6LoWPAN can also enhance sensor accuracy and timeliness.
- 3. Thin, soft-skin bioelectronics are compliant and can withstand high-cycle fatigue tests with repetitive mechanical deformations. However, their durability under long-term, dynamic skin movements such as stretching, compressing, and twisting requires robust adhesion and device configuration. Developing ultrathin biocompatible adhesives with good mechanical robustness is one approach. Still, these devices often suffer from limited stretchability and mechanical durability due to the trade-off between thinness and robustness. A thorough understanding and investigation of materials and rigid-soft interface design are necessary to prevent interface failure over long-term use.
- 4. Ensuring biofouling safety, longevity, and prevention is critical for overall performance and practicality. Wearable sensing devices must undergo rigorous testing to ensure stability and detection accuracy. Surface contamination, which reduces detection accuracy, poses a significant challenge. We should strengthen anti-fouling and effective calibration mechanisms to ensure stability during long-term use. Nonspecific binding of proteins, cells, or macromolecules can cause bioaccumulation, which can lead to sensing drift, especially in wearable oral biosensors operating in high protein environments like saliva. Protective coatings can minimize the impact of fouling and eliminate interference from electroactive substances.
- 5. To manage overall costs for successfully commercializing functionalized 2D materials-based biosensors, optimizing production processes, exploring specialized equipment and personnel, and utilizing cost-effective materials are essential. Additionally, joint efforts from materials and device engineers, data scientists, and medical professionals are required for technical progress. Closer involvement of users and caregivers is necessary before addressing common biosensor design issues such as bioreceptor selection, material biocompatibility, and sample matrix challenges. Sensor design teams must prioritize understanding the performance requirements of various users, including food consumers, environmental monitors, citizens using public services, healthcare workers, and regulatory agencies. Establishing key performance indicators in advance is vital to ensuring the biosensor's performance aligns with the desired quality of service (QoS) based on biosensor data.

In summary, integrating polymer-functionalized 2D materials with flexible circuit systems yields bioelectronic devices with versatile sensing capabilities, making them valuable for human body signal detection. Technological advancement and material performance are expected to enhance these devices, enabling high-quality disease monitoring. A profound understanding of emerging polymer-based 2D materials and innovative technologies promises to achieve biocompatible and highly effective bioelectronics and biosensors. It is expected that 2D nanomaterials will soon progress to the stage of clinical research and practical application.

7. Conclusion

The exploration of bioelectronics traces its roots back to the 18th century when Luigi Galvani conducted experiments on the electrical response of detached frog legs. On the other hand, research on 2D materials is still in its early stages, presenting opportunities and challenges for their integration into bioelectronic devices. This review delves into the recent advancements in utilizing 2D materials in bioelectronics,

examining the progress in fundamental research and device development. 2D materials, owing to their atomic thinness, high transparency, large specific area, distinctive optoelectrical properties, and biocompatibility, play a crucial role in the construction of biosensors and wearable bioelectronic devices. However, addressing significant challenges, particularly those associated with using 2D nanomaterials in biomedical science, is crucial before their therapeutic application. Key biomedical challenges include biological compatibility, adaptability, functionality, biocompatibility, and nano bio-surface characteristics. Polymers and 2D nanomaterials are essential to addressing biological issues within the human body. Polymer-functionalized 2D materials are more pliable and chemically attack-resistant than other 2D materials, offering advantages. Polymers can be easily functionalized or combined to form composites through covalent and non-covalent functionalization methods, surpassing conventional 2D materials. After the literature review, the authors can suggest that polymers are a promising choice for enhancing the traditional performance of 2D materials in biosensors and bioelectronics. This review paper closely examines how biocompatible polymers are added to the surfaces of 2D nanomaterials like g-C3N4, the graphene family, MXene, BP, MOF, and TMDCs. It focuses on their current state, physicochemical structures, synthesis methods, material properties, and uses in biosensors and bioelectronics.

CRediT authorship contribution statement

Tahreem Zahra: Writing – original draft, Conceptualization. Umme Javeria: Writing – review & editing, Conceptualization. Hasan Jamal: Writing – review & editing. Mirza Mahmood Baig: Writing – review & editing. Farid Akhtar: Writing – review & editing, Supervision. Urooj Kamran: Writing – review & editing, Supervision, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

No data was used for the research described in the article.

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